



**2011**  
**CARLYLE LAKE**  
**WATER QUALITY**  
**REPORT**



U.S. ARMY CORPS OF ENGINEERS, ST. LOUIS DISTRICT  
ENVIRONMENTAL ENGINEERING SECTION – WATER QUALITY

## Table of Contents

### Water Quality Report-Carlisle Lake

#### Section and Page No.

1.0 GENERAL OVERVIEW.....	pg. 1
2.0 WATER QUALITY ASSESSMENT CRITERIA.....	pg. 4
3.0 SUMMARY OF MONITORING RESULTS.....	pg. 9
4.0 PLANNED 2012 STUDIES .....	pg. 12

#### List of Figures & Tables

Figure 1: Lake Map.....	pg. 3
Table 2.1 State of Illinois - Water Quality Standards .....	pg. 4

#### Appendix

Appendix A: Data .....	pg. A1-A7
Appendix B: Laboratory Data Graphs	
E. Coli .....	pg. B1
Metals (Iron & Manganese) .....	pg. B2
Nitrogen (NH <sub>3</sub> -N & NO <sub>3</sub> -NO <sub>2</sub> ).....	pg. B3
Orthophosphate & Total Phosphate .....	pg. B4
Chlorophyll & Pheophytin .....	pg. B5
Pesticides.....	pg. B6
TSS & VSS .....	pg. B7
TOC.....	pg. B8
Appendix C: Field Data Graphs	
Temperature & DO .....	pg. C1-C3
Redox & Conductivity .....	pg. C4-C6
pH.....	pg. C7-C9
Secchi .....	pg. C10
2011 TW DO.....	pg. C11
Appendix D: Beach Water Quality Data & Graphs	
Keyesport .....	pg. D1

McNair .....	pg. D2
Coles Creek.....	pg. D3
Dam West.....	pg. D4

**Executive Summary**

The purpose of this report is to provide an annual analysis of the water quality in the lake for the past year. Carlyle Lake is located in south central Illinois, approximately 50 miles east of St. Louis, Missouri. The lake is 12 miles long and is 1 to 3 miles wide and has approximately 26,000 acres of water surface at summer pool. The lake shoreline is 83 miles and there are approximately 11,000 acres of public land managed primarily by the Corps of Engineers, but the Illinois Department of Natural Resources (IDNR) also plays a significant role in management of public lands. The lake is located on the Kaskaskia River at river mile 94.2 upstream from its confluence with the Mississippi River and approximately ½ mile upstream of the town of Carlyle.

The water of Carlyle Lake and the downstream river channel is generally good. The lake is a shallow reservoir susceptible to high winds. These conditions prevent the lake from stratifying permanently during the summer months. During low discharge during the heat of the summer fish kills have occurred in the old river channel below the dam due to low dissolved oxygen levels. Several years ago a remote sensor was installed on the spillway wall to allow project as well as water quality personnel to remotely monitor temperature and oxygen readings to avoid such fish kills by changing the release rate. No fish kills were observed during this past year.

All sampling sites met the appropriate state standards during 2011 except one E. coli reading and the phosphorous levels. The E. coli reading at Boulder Marina exceeded the state standard in April. No beach samples exceeded the state standard. Phosphorous levels have exceeded the state standard on a routine basis. Generally the tailwater and lake phosphorous levels are lower than the incoming tributary flows, which indicates that the lake is sinking the phosphorous. This is also occurring with nitrogen. The project area has several pollution potentials, with agriculture probably being the major contributor, but at present time, no major form of degradation to the lake or streams is apparent. Constant water quality monitoring will continue to check future degradation of the watershed.

## WATER QUALITY MONITORING PROGRAM

### **1.0 GENERAL OVERVIEW**

This report summarizes water quality activities of the St. Louis District for Fiscal Year 2011 in accordance with ER 1110-2-8154 Water Quality & Environmental management for Corps Civil Works Projects and ETL 1110-2-362 Environmental Engineering Initiatives for Water Management.

Water quality monitoring remains one of the Sections major responsibilities in the area of environmental stewardship. The objective is to maintain a reasonable environmental monitoring program for the Mississippi River and the 5 lakes under the St. Louis District's control. The District's reservoirs consist of Mark Twain and Wappapello Lakes in Missouri, and Shelbyville, Carlyle and Rend Lakes in Illinois. Water quality sampling is conducted within the lakes and their tributaries to establish trend analysis and maintain water quality at or above state and federal regulations.

The main objective is to provide technical expertise of an environmental nature to all Corps elements requesting assistance in accordance with ER 1110-2-8154. This would include updating the water quality management priorities for the district's projects to ensure water quality meets the state and federal regulations, for protection of human health and the environment, and for the safety and economic welfare of those at Corps projects. Ongoing goals include ensuring that downstream water quality meets all state and federal regulations, is suitable for aquatic and human life, and continue to evaluate trend analysis in relation to baseline conditions at all projects.

Water quality data is provided to the Illinois Environmental Protection Agency (IEPA) to be used in the Illinois Integrated Water Quality Report which is required every two years by the Clean Water Act Sections 303(d) and 305(b). IEPA does not typically monitor the three Corps lakes in Illinois. However, IEPA has stated that since the Corps lakes are the 3 largest lakes in the state, it is critical that their quality be routinely assessed. The state indicated that having the federally collected water quality data available now and in the future is critical to the state of Illinois meeting their mission in complying with the Clean Water Act Sections 305(b) and 303(d).

The National Water Quality Inventory Report to Congress 305(b) report is the primary vehicle for informing Congress and the public about general water quality conditions in the United States. This document characterizes our water quality, identifies widespread water quality problems of national significance, and describes various programs implemented to restore and protect our waters.

Under Section 303(d) of the 1972 Clean Water Act, states, territories and authorized tribes are required to develop a list of water quality limited segments. These waters on the list do not meet water quality standards, even after point sources of pollution have installed the minimum required levels of pollution control technology. The law requires that these jurisdictions establish priority rankings for water on the lists and develop action plans, called as Total Maximum Daily Loads (TMDL), to improve water quality.

Currently the Illinois Environmental Protection Agency (IEPA) has listed Carlyle Lake impaired for Total Suspended Solids (TSS), Total Phosphorous, mercury, and manganese. The lists of sources for these impairments are contaminated sediments, crop production, and unknown sources. The Kaskaskia River is impaired by the above parameters and also Fecal Coliform and Polychlorinated biphenyls. Continued monitoring of the lake and its tributaries is vital in assisting the future assessment of the lake for these and other possible impairments. The water quality monitoring program represents the single metric that encompasses the overall health of the watershed as it is a direct measure of how well the environmental stewardship programs are working.

## **1.1 INTRODUCTION**

Carlyle Lake is within the Kaskaskia River Basin in central Illinois. The lake serves as a heavy recreational usage lake and supplies water to numerous communities. The land surrounding the lake is used predominately for agriculture. Surrounding communities have existing industrial/commercial operations as well as residential communities which discharge wastewater into municipal wastewater treatment plants which ultimately discharge treated water into the Kaskaskia River Basin. Agricultural runoff and municipal wastewater treatment facilities are the primary potential source of pollution into the Carlyle Lake watershed. Additional sources are marinas, recreational watercraft discharges and effluent from nearby subdivisions and a golf course.

Water quality monitoring was conducted during 2011 to assure safe conditions for human recreation, wildlife and aquatic life as maintained and managed within the lake system. The 2011 water quality monitoring program was substantially reduced due to reduced funding. Previous to 2009 five sampling events were conducted during the recreational season. In the initial phase of the sampling program during the 1970's and 80's six or seven sampling events were conducted. A restored number of sampling events would provide the ability to better evaluate water quality trends, to better defend project operations (lake levels, releases, maintenance projects, construction projects, etc.), to better confirm that we meet state water quality standards, and to better confirm that human health and safety are adequately protected. Four sampling events were conducted at 5 sites. During the sampling period one site was selected for quality control duplication and denoted as CAR-15. The locations of the five sampling sites are depicted on the lake map in Figure 1.

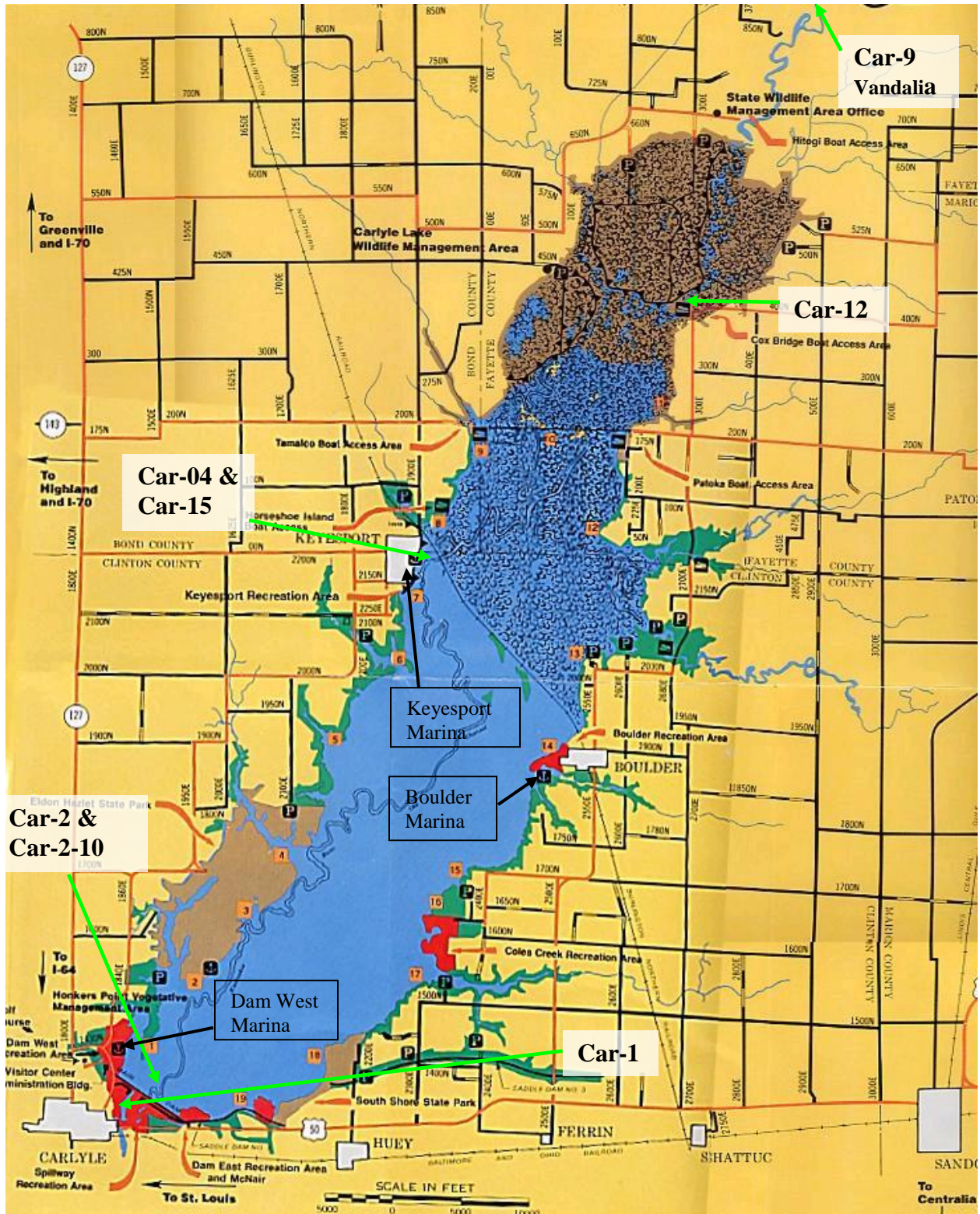


Figure 1  
Location of sample sites

## 2.0 WATER QUALITY ASSESSMENT CRITERIA

The water quality assessment criteria were based upon the State of Illinois regulatory limits for certain contaminants, which has been generally accepted criteria for sustaining adequate aquatic plant and animal growth. The samplings and analysis which were conducted at the Carlyle Lake sites reflect the minimal set of parameters needed to analyze the current status of water quality for the Carlyle Lake system. In addition to the water quality sampling, sediment samples were also taken at the lake sites. Sediment samples were analyzed for metals and semi-volatile organophosphates.

The following parameters were analyzed in the Fiscal Year 2011 samplings at Carlyle Lake: Total Organic Carbon (TOC), iron, manganese, ammonia-nitrogen, nitrate-nitrogen, orthophosphate, total phosphate, Total Suspended Solids (TSS), Total Volatile Suspended Solids (TVSS), fecal coliform, pH, temperature, dissolved oxygen, specific conductance, oxidation-reduction potential (ORP), chlorophyll, pheophytin-a, atrazine and alachlor.

### 2.1 WATER

The Illinois Environmental Protection Agency in Title 35, Subtitle, C, classifies water quality criteria based on end usage. Subpart B contains regulations for general use water, while subparts C and D delineate those for public and food processing water and secondary contact and indigenous aquatic life standards, respectively. These standards are used to determine the aquatic water quality of the lake. Table 2.1 provides a listing of the regulatory limits for the parameters analyzed where a limit has been established.

<b>TABLE 2.1</b>	
<b>State of Illinois</b>	
<b>Water Quality Standards</b>	
<b>PARAMETER</b>	<b>LIMIT</b>
Temperature	Rise of 2.8°C above normal seasonal temp
Ammonia Nitrogen	15 mg/L
Nitrate Nitrogen	10 mg/L
Total Iron	2.0 mg/L (2 <sup>nd</sup> Contact & Aquatic Life)
Manganese	1.0 mg/L
Total Phosphate	0.05 mg/L Lakes; 0.61 mg/L Streams
E. Coli	Illinois standard is 235 E. coli per 100ml for single sample or 126 for geometric mean.
pH	Range: 6.5 to 9.0
DO	> 5.0 mg/L
Conductivity	1,667 $\mu$ S/cm $\approx$ TDS of 1,000 mg/L
Total Suspended Solids (TSS)	116mg/L (Streams); $\geq$ 12mg/L (Lakes)
Atrazine	0.003 mg/L <sup>1</sup> ; 82 $\mu$ g/L <sup>2</sup> ; 9 $\mu$ g/L <sup>3</sup>
Alachlor	0.002 mg/L (Drinking Water Standard)
Cyanazine	370 $\mu$ g/L Acute; 30 $\mu$ g/L <sup>3</sup>
Metolachlor	1.7mg/L <sup>2</sup>
Simazine	4.0 $\mu$ g/L <sup>1</sup>

Trifluralin	26ug/L Acute; 1.1ug/L <sup>3</sup>
Pendmethalin (PROWL)	70ug/L HBSL, 20ug/L <sup>1</sup>

<sup>1</sup> Drinking Water Standard

<sup>2</sup> Acute

<sup>3</sup> Chronic

Health Based Screening Level (HBSL)

Nitrogen is an essential component of proteins, genetic material, chlorophyll, and other key organic molecules. All organisms require nitrogen in order to survive. Nitrogen exists in several forms. These forms include gaseous nitrogen (N<sub>2</sub>), nitrites (NO<sub>2</sub>), nitrate (NO<sub>3</sub>), ammonia nitrogen (NH<sub>3</sub>-N), and ammonium (NH<sub>4</sub>). Ammonia can be toxic to fish and other aquatic organisms at certain levels. Unlike ammonia, ammonium (NH<sub>4</sub>) is not toxic to aquatic organisms and is readily available for uptake by plankton and macrophytes. Nitrogen levels have increased as human activities have accelerated the rate of fixed nitrogen being put into circulation. High nitrogen levels can cause eutrophication. Eutrophication increases biomass of phytoplankton, decrease water transparency, and causes oxygen depletion. Ammonia nitrogen is monitored so that the effects on fish spawning, hatching, growth rate and pathologic changes in gills, liver and kidney tissue can be related to the detected levels of ammonia nitrogen. Nitrate-nitrogen degrades to nitrite or produces ammonia which has a detrimental effect on aquatic life and, therefore, has been monitored to assure levels are below the regulatory "safe" limit.

Phosphate has been analyzed as phosphorus and has been monitored due to the potential for uptake by nuisance algae. Levels of phosphate can indicate the potential for rapid growth of algae (algae bloom) which can cause serious oxygen depletion during the algae decay process. Phosphorous is typically the limiting nutrient in a water body. Therefore, addition of phosphorous to the ecosystem stimulates the growth of plants and algae. Phosphorous is delivered to lakes and streams by way of storm water runoff from agricultural fields, residential property, and construction sites. Other sources of phosphorous are anaerobic decomposition of organic matter, leaking sewer systems, waterfowl, and point source pollution. The general standard for phosphorous in lake water is 0.05mg/L. Dissolved phosphorous also called ortho-phosphorous is generally found in much smaller concentrations than total phosphorous and is readily available for uptake. For this reason dissolved phosphorous concentrations are variable and difficult to use as an indicator of nutrient availability.

The metals manganese and iron are nutrients for both plants and animals. Living organisms require trace amounts of metals. However, excessive amounts can be harmful to the organism. Heavy metals exist in surface waters in three forms, colloidal, particulate, and dissolved. Water chemistry determines the rate of adsorption and desorption of metals to and from sediment. Metals are desorbed from the sediment if the water experiences increases in salinity, decreases in redox potential, or decreases in pH. Metals in surface waters can be from natural or human sources. Currently human sources contribute more metals than natural sources. Metal levels in surface water may pose a health risk to humans and the environment.

Photosynthetic activity can be hindered by the levels of total suspended solids. Total suspended solids concentrations, which cause the photosynthetic activity to be reduced by more than 10% from the seasonably established norm, can have a detrimental effect on aquatic life. Soil particles, organic material, and other debris comprise suspended solids in the water column. Secchi



disk measurements are inverse to suspended solid measurements. As the total suspended solids (TSS) increase, the secchi disk depth or water transparency decreases. Total suspended solids can be an important indicator of the type and degree of turbidity. TSS measurements represent a combination of volatile suspended solids (VSS) which is comprised of organic material and nonvolatile suspended solids (NVSS) which is comprised of inorganic mineral particles in the water. In order to more accurately determine the types and amounts of suspended solids, volatile suspended solids (VSS) are analyzed. VSS concentration represents the organic portion of the total suspended solids. Organic material often includes plankton and additional plant and animal debris that is present in water. Total volatile solids indicate the presence of organics in suspension and, therefore, show additional demand levels of oxygen. Illinois does not currently have a standard for TSS or TVSS. However, literature suggests that NVSS above 15mg/L could highly impair recreational lake use and a NVSS of 3 to 7mg/L might cause slight impairment.

Chlorophyll and pheophytin-a are monitored to provide indicators of algae growth and, therefore, potential oxygen depletion activity. Chlorophyll is measured in lakes to estimate the type and amount of algal productivity in the water column. Chlorophyll a is present in green algae, blue-green algae, and in diatoms. Chlorophyll a is often used to indicate the degree of eutrophication. Chlorophyll b and c are used to estimate the extent of algal diversity and productivity. Chlorophyll b is common in green algae and is used as an auxiliary pigment for photosynthesis. Chlorophyll c is most common in diatom species and serves as an auxiliary pigment. Algal productivity and diversity can be determined by the concentrations of the individual pigments. For example high concentrations of chlorophyll a and b would indicate that green algae is abundant. High concentrations of chlorophyll a would indicate abundance of blue-green algae and concentrations of chlorophyll a and c would indicate diatoms are the dominant species. Chlorophyll production is currently being connected with hypoxia.

Fecal coliform bacteria is monitored for the protection of human health as it relates to full body contact of recreational waters. People can be exposed to disease-causing organisms, such as bacteria, viruses and protozoa in beach and recreational waters mainly through accidental ingestion of contaminated water or through skin contact. These organisms, called pathogens, usually come from the feces of humans and other warm-blooded animals. If taken into the body, pathogens can cause various illnesses and on rare occasions, even death. Waterborne illnesses include diseases resulting from bacteria infection such as cholera, salmonellosis, and gastroenteritis, viral infections such as hepatitis, gastroenteritis, and intestinal diseases, and protozoan infections such as amebic dysentery and giardiasis. The most commonly monitored recreational water indicator organisms are fecal coliform, *Escherichia coli*, (*E. coli*) and enterococci. Fecal coliform are bacteria that live in the intestinal tracts of warm-blooded animals. The standard for fecal coliform is less than 500 colonies per 100ml of sample water. Fecal coliform was originally recommended in 1968 by the Federal Water Pollution Control Administration (predecessor to EPA) as an effective water quality indicator organism for recreational waters. Recent studies indicate that fecal coliform show less correlation to illness than other indicator organisms such as *E. coli* and enterococci. The Environmental Protection Agency (EPA) currently recommends *E. coli* or enterococci as an indicator organism for fresh waters. Since 2009 the St. Louis District has been using *E. coli* as the standard indicator.

Atrazine and Alachlor herbicides are commonly used agricultural chemicals which can be readily transported by rainfall runoff. Both compounds are suspected of causing cancer and, therefore,

were monitored for the protection of human and aquatic health. Organic compounds include many pesticides. A pesticide can be any substance that is intended to prevent, destroy, repel, or mitigate any pest. This includes insecticides, herbicides, fungicides, fumigants, algaecides and other substances. Herbicides which are pesticides used to kill vegetation are the most widely used and sampled. Ten of the most frequently used herbicides and detected in water are Atrazine, Metolachlor, Alachlor, 2,4-D, Trifluralin, Glyphosate, Dicamba, Cyanazine, Simazine, and 2,4,5-T. Two of the most widely used pesticides are Atrazine and Alachlor. Atrazine is a preemergence or postemergence herbicide use to control broadleaf weeds and annual grasses. Atrazine is most commonly detected in ground and surface water due to its wide use, and its ability to persist in soil and move in water. Alachlor is a Restricted Use Pesticide (RUP) due to the potential to contaminate groundwater. The drinking water standard for Atrazine is 0.003mg/L and 0.002 mg/L for Alachlor.

Temperature, dissolved oxygen and pH are monitored for the protection of aquatic life. Temperature is important because it controls several aspects of water quality. Colder water hold more dissolved oxygen which is required by aquatic organisms. Plants grow more rapidly and use more oxygen in warmer water. Decomposition of organic matter which uses oxygen is accelerated in warmer water. Temperature can also determine the availability of toxic compounds such as ammonia. Since aquatic organisms are cold blooded, water temperature regulates their metabolism and ability to survive. The number and kinds of organisms that are found in streams or lakes is directly related to temperature. Certain organisms require a specific temperature range, such as trout, which require water temperatures below 20°C. Most aquatic organisms require a minimum concentration of dissolved oxygen to survive. In spring, surface waters of the lake mix with the water below through wind and thermal action. This mixing diminishes as the upper layer of water becomes warmer and less dense. Solar insolation during the summer months stratifies the lake into three zones. The upper warmer water zone is called the epilimnion and the lower cooler water zone is called the hypolimnion. The epilimnion and the hypolimnion zones are divided by a transition zone known as the metalimnion. The thermocline located within the metalimnion exhibits a rapid change in water temperature. During the summer months the hypolimnion may become anaerobic. In this anaerobic zone, chemical reduction of iron and manganese, or the production of methane and sulfides can occur. Iron rapidly oxidizes in aerobic environments, but manganese oxidizes slowly and can remain in the reduced state for long distances down stream even in aerobic environments. The degree of acidity of water is measured by a logarithmic scale ranging from 0 to 14 and is known as the pH scale. A reading of 7 indicates neutrality and readings below seven are acidic and above are alkaline. Most Illinois lakes range from 6 to 9 on the pH scale. The buffering capacity of water is the ability to neutralize acid better known as alkalinity. A high alkalinity concentration indicates an increased ability to neutralize pH and resist changes, whereas a low alkalinity concentration indicates that a water body is vulnerable to changes in pH.

Conductivity is a measure of a water's ability to conduct an electrical current. The ability to carry a current is often driven by the dissolved materials present in a water column. These materials can include dissolved ions and other materials in the water and thus are directly proportional to the concentration of total dissolved solids (TDS) present in the water column. Typically TDS concentrations represent 50-60% of the conductivity measurements. Conductivity is also affected by water temperature. The warmer the water, the higher the conductivity. Conductivity in streams and rivers is affected by the geology of the area. Streams running through granite areas tend to have lower conductivity due to granite being composed of inert material, materials that do not ionize or dissolve

into ionic compounds in water. On the other hand streams that run through areas of limestone or clay soils tend to have higher conductivity readings because of the presence of materials that ionize. Conductivity is useful as a general measure of water quality. A stream tends to have a relatively constant range of conductivity that once established can be used as a baseline. Significant changes either high or low might indicate a source of pollution has been introduced into the water. The pollution source could be a treatment plant which raises the conductivity or an oil spill which would lower the conductivity.

Redox or Oxidation-Reduction Potential (ORP) is a measurement to oxidize materials. Oxidation involves an exchange of electrons between 2 atoms. The atom that loses an electron is oxidized and the one that gains an electron is reduced. ORP sensors measure the electrochemical potential between the solution and a reference electrode. Readings are expressed in millivolts with positive readings indicating increased oxidizing potential and negative readings being increased reduction. The ORP probe is essentially a millivolt meter, measuring the voltage across 2 electrodes with the water in between. ORP values are used much like pH values to determine water quality. While pH readings characterize the state of a system relative to the receiving or donating hydrogen ions (base or acid), ORP readings characterize the relative state of losing or gaining electrons. The conversion of ammonia ( $\text{NH}_3$ ) requires an oxidizing environment to convert it into nitrites ( $\text{NO}_2$ ) and nitrates ( $\text{NO}_3$ ). Ammonia levels as low as 0.002mg/L can be harmful to fish. Generally ORP readings above 400mV are harmful to aquatic life. However, ORP is a non-specific measurement which is a reflection of a combination of effects of all the dissolved materials in the water. Therefore, the measurement of ORP in relatively clean water has only limited utility unless a predominant redox-active material is known to be present.

Water clarity is intuitively used by the public to judge water quality. Secchi depth has been used for many years as a limnological characterization tool for characterizing water clarity. Secchi depth is a measure of light penetration into a waterbody and is a function of the absorption and scattering of light in the water. There are three characteristics of water which affect the penetration of light. The three factors are the color of water, amount of phytoplankton in the water column, and amount of inorganic material in the water column. Secchi depth integrates the combined impacts of all the factors which influence water clarity. Water transparency was measured using a Secchi disk. Secchi disk readings were taken at all lake sites.

## **2.2 Sediment**

In accordance with EM-1110-2-1201, sediment samples should be taken to monitor and assess potential impacts to aquatic and human health. To assess ecological risk, sample values were compared against toxicity information published in the National Oceanic Atmospheric Administrations (NOAA) Screening Quick Reference Tables (SQRT) or similar references for ecological receptors in freshwater sediment. Without standards or other widely applicable numerical tools, NOAA scientists found it difficult to estimate the possible toxicological significance of chemical concentrations in sediment. Therefore, numerical sediment quality guidelines (SQG's) were developed as informal, interpretive tools. The SQGs were not promulgated as regulatory standards, but rather as informal, non-regulatory guidelines for interpreting chemical data from analyses of sediments. For potential ecological risk from inorganic contaminants, seven metals are typically of "most concern" with regards to fish and wildlife: Arsenic, Copper, Cadmium, Selenium, Mercury, Lead, and Zinc. Avian species are thought to be particularly sensitive to arsenic, but is also considered a carcinogenic, mutagenic, and

teratogenic contaminant in a variety of species in elevated doses over time. Avian species are also known to be particularly sensitive to lead in the environment with effects ranging from mortality, reduced growth and reproductive output, behavior changes, blood chemistry alterations, and lesions of major organs. Finally, the embryo stages in fish and avian species are known to be the most sensitive life stage to selenium effecting reproductive success.

It is recommended that the next round of sediment samples focus on organochlorines in freshwater sediment to assess potential chronic aquatic impacts (e.g. aldrin, chlordane, endrin, endosulfan, DDT, methoxychlor).

For potential human health risk, there are no known values in Illinois for sediments. While not a direct correlation, sample results were compared against Illinois Tiered Approach to Corrective Action Objectives (TACO) and Non-TACO lowest default target levels for all soil types and exposure pathways for soils.

### **3.0 SUMMARY OF MONITORING RESULTS**

#### **3.1 Water Quality Summary**

The monitoring program for Carlyle Lake during Fiscal Year 2011 revealed good water quality when compared to limits established by the Illinois Environmental Protection Agency (IEPA) for general use, secondary contact, and indigenous aquatic life. Normally seasonal change brings on gradual lake stratification during the summer months. Water quality trends on a yearly basis are hard to determine when only conducting 4 sampling events. However, over the course of a 5 year period these 4 sampling events per year are adequate to determine trends in water quality. Agricultural nutrient runoffs were primary concerns for the lake's water quality. Better land management practices, erosion control and buffering zones are methods used to reduce such contaminants from entering the lake. The St. Louis District Environmental Engineering Section participates in the annual Kaskaskia Watershed Summit which provides information about the condition of the entire watershed.

E. coli are sampled at the marinas to ensure that the marina areas are not being contaminated by boats with restroom facilities. All E. coli levels at both marinas were below the Illinois standard of 235 mpn/100ml except the April 27 sample at Boulder, which was 789. This may have been due to the number of gulls in the area and the 2.21 inches of rain that fell in the area over the previous 48 hours. The results of the samples are not received for 30 days after the samples are taken. The project office was not notified of the result because of the delay in obtaining the results, fishing and swimming are not allowed in the marina area, and samples are taken to determine if there is a re-occurring issue. In the future, the project office will be notified as soon as any readings not meeting standards are received. E. coli beach data was received from the project office. No beach samples were above the Illinois standard.

Total iron and total manganese are sampled above the dam near the bottom of the channel (Car-2-10) and in the spillway area (Car-1). As was previously stated living organisms require trace amounts of metals, however excessive amounts can be harmful to the organism. Iron did not exceed the Illinois Water Quality Standard at either site Car-1 or Car-2-10. Iron cycling is a function of oxidation-reduction processes. This elevated level of iron near the bottom of the lake is not

detrimental to the overall lake system at this time. Iron oxidizes relatively rapidly (minutes to hours); therefore any iron released through the spillway will be oxidized in a short period of time. Manganese did not exceed the Illinois water quality general use standard of 1.0 mg/L at either site. Illinois does not currently have a general use standard for iron.

Nitrogen and phosphates are sampled at all sites. As for the past several years the 2011 phosphate results at all sites are above the 0.05 mg/L standard. Because phosphorus in water is not considered directly toxic to humans and animals no drinking water standards have been established for phosphorus. However, phosphorus can cause health threats through the stimulation of toxic algal blooms and the resulting oxygen depletion. However, nitrates can pose a threat to human and animal health. Nitrate in water is toxic at high levels and has been linked to toxic effects of livestock and to blue baby disease (methemoglobinemia) in infants. The Maximum Contaminant Level (MCL) for nitrate-N in drinking water is 10mg/L to protect babies 3 to 6 months of age. The Illinois Water Quality Standard for ammonia nitrogen (NH<sub>3</sub>-N) is 15mg/L. The increased levels of phosphate in combination with nitrogen and other lake conditions, such as temperature, pH and stagnant lake conditions, can lead to increased algae growth. Eutrophication is currently the most widespread water quality problem in the U.S. and many other countries. Restoration of eutrophic waters requires the reduction of nonpoint inputs of phosphorous and nitrogen. The resulting detrimental effects of algae toxins and oxygen depletion could result in health problems for fish and other aquatic species as well as land animals utilizing the water supply. There were no signs of any of these effects throughout 2011. Both Nitrate-Nitrogen and Ammonia-Nitrogen decreased as water transverses down the lake. The lake appears to capture and use up nitrogen which reduces nutrient levels released from the lake. This reduction of nutrient levels traveling down stream results in an improvement of water quality. The phosphates however, appeared to accumulate lower in the lake. Agriculture in the area probably is a major contributor to these results.

Chlorophyll *a* was sampled at 3 sites, Car-2, Car-4, (Car-15 is a duplicate sample at site Car-4) and Car-12. . Illinois does not currently have a standard for chlorophyll. The data indicates a normal increase in chlorophyll levels during the warmer summer months, which is not a concern. Chlorophyll *a* is a green pigment found in plants. Chlorophyll *a* concentrations are an indicator of phytoplankton abundance and biomass. They can be an effective measure of trophic status, and used as a measure of water quality. High levels often indicate poor water quality and low levels suggest good conditions. However, elevated levels are not necessarily bad. It is the long term persistence of elevated levels that is the problem. It is natural for chlorophyll *a* levels to fluctuate over time. Chlorophyll *a* tends to be higher after storm events and during the summer months when water temperatures and light levels are elevated. Chlorophyll can reduce the clarity of the water and the amount of oxygen available to other organisms. Illinois does not currently have a standard for chlorophyll. The data indicates a normal increase in chlorophyll levels during the warmer summer months, which is not a concern.

Atrazine and Alachlor are pesticides that were sampled at all sites. These chemicals are herbicides used to control weed growth. Neither Atrazine nor Alachlor exceeded the Illinois drinking water standards in 2011. Cyanazine, Metolachlor, Trifluralin and Simazine are also analyzed as part of the pesticide screening. None of these constituents exceeded Illinois standards. The Carlyle Lake watershed consists of approximately 75% cropland. These substances can enter water bodies as a result of drift during spraying, surface runoff, and leaching through soil. In order to eliminate pesticide contamination of waters it is important for the public to be educated and institute best manage practices

when using these chemicals.

Total Suspended Solids (TSS) and Total Volatile Suspended Solids (TVSS) samples are collected at all sites. CAR-9 exceeded the TSS Illinois standard for streams. All lake sites exceeded the Illinois standard for TSS in lakes. Solids can affect water quality by increasing temperature through the absorption of sunlight by the particles in the water, which also effects the clarity of the water. This can then effect the amount of oxygen in the water. Illinois guidelines for indentifying potential causes of impairment of aquatice life in lakes list a TSS above 12mg/L could impair recreational lake use and a TSS of 116mg/L may cause impairment of streams. Data indicates that sediment settles out as it travels down the lake.

Total Organic Carbon (TOC) is collected at all sites. Data indicates that early in the year TOC is higher in the upper portions of the lake. TOC is an indicator of the organic character of water. The larger the carbon or organic content, the more oxygen is consumed. This may be a result of plant material, which had grown all summer and begins to decay. Illinois does not currently have a standard for TOC. Since Illinois does not have a standard for this parameter, observations of high or low are relative to the current sampling period.

Temperature and dissolved oxygen levels were taken at all sites. See Appendix C graphs C1-C3. Measurements were taken at 1 meter intervals at the lake sites. During the summer months the lake stratifies and a boundary is formed between the upper warmer water and the lower cooler water. This transition area is known as the thermocline, the area where the temperature drops significantly. Oxygen levels can also change drastically as a function of depth. This area where the oxygen level significantly drops is called the oxycline. The depth of the thermocline and oxycline can have an effect on the aquatic organisms. Occasionally the thermocline and oxycline are at or near the same depth.

pH is taken at all sites and at 1 meter intervals at lake sites. All sites were within the Illinois standard of 6 to 9 pH range. Variances in pH can be caused by increase runoff due to a rainfall event, unusual temperature extremes, or erosion from land disturbances. Another cause may be that photosynthesis uses up dissolved carbon dioxide, which acts like carbonic acid ( $H_2CO_3$ ) in water.  $CO_2$  removal in effect reduces the acidity of the water thus the pH increases.

Conductivity and redox are taken at all sites and at 1 meter intervals at lake sites. Illinois does not currently have a standard for redox, but does have a standard of less than 1,667 uS/cm for conductivity. No samples exceeded this standard.

Seechi disk readings indicate that as the water travels down the lake it becomes clearer. This is most likely the result of sediments dropping out of the water column as the water moves down stream toward the dam.

The remote sensor in the spillway was monitored and maintained throughout the year to allow the project as well as water quality personnel to remotely monitor temperature and oxygen readings to acquire data to inform operational actions in order to avoid fish kills. During low flow, water is discharged through the sluice gates from the bottom of the lake. This water is low in oxygen and can create a low oxygen area below the dam. The sensor allows the project to track oxygen levels below

the dam and make appropriate adjustments to avoid a possible fish kill. Normally allowing water to spill through the tainter gates will alleviate low oxygen levels below the dam. No fish kills were observed this year. The sonde was serviced approximately once each month from May through September. Dissolved oxygen did drop below the 5mg/l standard in July, August, and September due to the algae build up on the probe.

### **3.2 Sediment Summary**

Sediment sampling was not conducted in 2011. Sediment sampling is normally conducted every 5 years if funding is available. Sediment sampling was last conducted in 2007.

### **4.0 PLANNED 2012 STUDIES**

The Carlyle Lake water quality monitoring will continue in Fiscal Year 2012 on a limited basis. Because of budgetary constraints there will only be 3 sampling events in 2012. Reduction of the number of sampling events results in the inability to evaluate water quality trends, the inability to scientifically defend operations, the inability to confirm state water quality standards, and the inability to adequately protect human health and safety. Carlyle Lake provides water supplies to many communities and is a high usage recreational lake. The monitoring of water quality is imperative to assure the water quality is within acceptable limits for the designated usage.

The sampling sites include the following: Site 1 Car-1 Spillway, Site 2 Car-02 Lake side in front of Dam, Site 4 Car-04 Keyesport, Site 9 Kaskaskia River at Vandalia, and Site 12 Cox Bridge. This combination of sites effectively represents the incoming contaminants and their effects on the lake.

In addition, water quality personnel will continue to maintain and remotely monitor the DO & temperature probe in the spillway.

## **APPENDIX A**

### **DATA LAB DATA**



## LAB DATA

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-1	4/27/2011	WT	ND		ug/L	Acetochlor
Car-1	6/23/2011	WT	ND		ug/L	Acetochlor
Car-1	8/25/2011	WT	ND		ug/L	Acetochlor
Car-12	4/27/2011	WT	ND		ug/L	Acetochlor
Car-12	6/23/2011	WT	ND		ug/L	Acetochlor
Car-12	8/25/2011	WT	ND		ug/L	Acetochlor
Car-15	4/27/2011	WT	ND		ug/L	Acetochlor
Car-15	6/23/2011	WT	ND		ug/L	Acetochlor
Car-2	4/27/2011	WT	ND		ug/L	Acetochlor
Car-2	6/23/2011	WT	ND		ug/L	Acetochlor
Car-2	8/25/2011	WT	ND		ug/L	Acetochlor
Car-4	4/27/2011	WT	ND		ug/L	Acetochlor
Car-4	6/23/2011	WT	ND		ug/L	Acetochlor
Car-4	8/25/2011	WT	ND		ug/L	Acetochlor
Car-9	4/27/2011	WT	ND		ug/L	Acetochlor
Car-9	6/23/2011	WT	ND		ug/L	Acetochlor
Car-9	8/25/2011	WT	ND		ug/L	Acetochlor
Car-1	4/27/2011	WT	ND		ug/L	Alachlor
Car-1	6/23/2011	WT	ND		ug/L	Alachlor
Car-1	8/25/2011	WT	ND		ug/L	Alachlor
Car-12	4/27/2011	WT	ND		ug/L	Alachlor
Car-12	6/23/2011	WT	ND		ug/L	Alachlor
Car-12	8/25/2011	WT	ND		ug/L	Alachlor
Car-15	4/27/2011	WT	ND		ug/L	Alachlor
Car-15	6/23/2011	WT	ND		ug/L	Alachlor
Car-2	4/27/2011	WT	ND		ug/L	Alachlor
Car-2	6/23/2011	WT	ND		ug/L	Alachlor
Car-2	8/25/2011	WT	ND		ug/L	Alachlor
Car-4	4/27/2011	WT	ND		ug/L	Alachlor
Car-4	6/23/2011	WT	ND		ug/L	Alachlor
Car-4	8/25/2011	WT	ND		ug/L	Alachlor
Car-9	4/27/2011	WT	ND		ug/L	Alachlor
Car-9	6/23/2011	WT	ND		ug/L	Alachlor
Car-9	8/25/2011	WT	ND		ug/L	Alachlor
Car-1	4/27/2011	WT	0.15		mg/L	Ammonia-N
Car-1	6/23/2011	WT	0.19		mg/L	Ammonia-N
Car-1	8/25/2011	WT	ND		mg/L	Ammonia-N
Car-12	4/27/2011	WT	0.28		mg/L	Ammonia-N
Car-12	6/23/2011	WT	0.20		mg/L	Ammonia-N
Car-12	8/25/2011	WT	ND		mg/L	Ammonia-N
Car-15	4/27/2011	WT	0.08	J	mg/L	Ammonia-N
Car-15	6/23/2011	WT	0.24		mg/L	Ammonia-N
Car-15	8/25/2011	WT	0.20		mg/L	Ammonia-N
Car-2	4/27/2011	WT	ND		mg/L	Ammonia-N
Car-2	6/23/2011	WT	0.20		mg/L	Ammonia-N
Car-2	8/25/2011	WT	ND		mg/L	Ammonia-N

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-2-10	4/27/2011	WT	ND		mg/L	Ammonia-N
Car-2-10	8/25/2011	WT	ND		mg/L	Ammonia-N
Car-4	4/27/2011	WT	0.15		mg/L	Ammonia-N
Car-4	6/23/2011	WT	0.22		mg/L	Ammonia-N
Car-4	8/25/2011	WT	0.19		mg/L	Ammonia-N
Car-9	4/27/2011	WT	0.09	J	mg/L	Ammonia-N
Car-9	6/23/2011	WT	0.15		mg/L	Ammonia-N
Car-9	8/25/2011	WT	ND		mg/L	Ammonia-N
Car-1	4/27/2011	WT	ND		ug/L	Atrazine
Car-1	6/23/2011	WT	ND		ug/L	Atrazine
Car-1	8/25/2011	WT	ND		ug/L	Atrazine
Car-12	4/27/2011	WT	ND		ug/L	Atrazine
Car-12	6/23/2011	WT	ND		ug/L	Atrazine
Car-12	8/25/2011	WT	ND		ug/L	Atrazine
Car-15	4/27/2011	WT	0.51	J	ug/L	Atrazine
Car-15	6/23/2011	WT	ND		ug/L	Atrazine
Car-2	4/27/2011	WT	ND		ug/L	Atrazine
Car-2	6/23/2011	WT	ND		ug/L	Atrazine
Car-2	8/25/2011	WT	ND		ug/L	Atrazine
Car-4	4/27/2011	WT	0.47	J	ug/L	Atrazine
Car-4	6/23/2011	WT	ND		ug/L	Atrazine
Car-4	8/25/2011	WT	ND		ug/L	Atrazine
Car-9	4/27/2011	WT	0.57	J	ug/L	Atrazine
Car-9	6/23/2011	WT	ND		ug/L	Atrazine
Car-9	8/25/2011	WT	ND		ug/L	Atrazine
Car-1	4/27/2011	WT	ND		ug/L	Bromacil
Car-1	6/23/2011	WT	ND		ug/L	Bromacil
Car-1	8/25/2011	WT	ND		ug/L	Bromacil
Car-12	4/27/2011	WT	ND		ug/L	Bromacil
Car-12	6/23/2011	WT	ND		ug/L	Bromacil
Car-12	8/25/2011	WT	1.10		ug/L	Bromacil
Car-15	4/27/2011	WT	ND		ug/L	Bromacil
Car-15	6/23/2011	WT	ND		ug/L	Bromacil
Car-2	4/27/2011	WT	ND		ug/L	Bromacil
Car-2	6/23/2011	WT	ND		ug/L	Bromacil
Car-2	8/25/2011	WT	ND		ug/L	Bromacil
Car-4	4/27/2011	WT	ND		ug/L	Bromacil
Car-4	6/23/2011	WT	ND		ug/L	Bromacil
Car-4	8/25/2011	WT	ND		ug/L	Bromacil
Car-9	4/27/2011	WT	ND		ug/L	Bromacil
Car-9	6/23/2011	WT	ND		ug/L	Bromacil
Car-9	8/25/2011	WT	ND		ug/L	Bromacil
Car-1	4/27/2011	WT	ND		ug/L	Butachlor
Car-1	6/23/2011	WT	ND		ug/L	Butachlor
Car-1	8/25/2011	WT	ND		ug/L	Butachlor
Car-12	4/27/2011	WT	ND		ug/L	Butachlor
Car-12	6/23/2011	WT	ND		ug/L	Butachlor
Car-12	8/25/2011	WT	ND		ug/L	Butachlor

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-15	4/27/2011	WT	ND		ug/L	Butachlor
Car-15	6/23/2011	WT	ND		ug/L	Butachlor
Car-2	4/27/2011	WT	ND		ug/L	Butachlor
Car-2	6/23/2011	WT	ND		ug/L	Butachlor
Car-2	8/25/2011	WT	ND		ug/L	Butachlor
Car-4	4/27/2011	WT	ND		ug/L	Butachlor
Car-4	6/23/2011	WT	ND		ug/L	Butachlor
Car-4	8/25/2011	WT	ND		ug/L	Butachlor
Car-9	4/27/2011	WT	ND		ug/L	Butachlor
Car-9	6/23/2011	WT	ND		ug/L	Butachlor
Car-9	8/25/2011	WT	ND		ug/L	Butachlor
Car-12	4/27/2011	WT	6.40		mg/m3	Chlorophyll a
Car-12	6/23/2011	WT	8.54		mg/m3	Chlorophyll a
Car-12	8/25/2011	WT	14.95		mg/m3	Chlorophyll a
Car-15	4/27/2011	WT	8.54		mg/m3	Chlorophyll a
Car-15	6/23/2011	WT	6.41		mg/m3	Chlorophyll a
Car-15	8/25/2011	WT	36.31		mg/m3	Chlorophyll a
Car-2	4/27/2011	WT	46.99		mg/m3	Chlorophyll a
Car-2	6/23/2011	WT	23.50		mg/m3	Chlorophyll a
Car-2	8/25/2011	WT	36.31		mg/m3	Chlorophyll a
Car-4	4/27/2011	WT	6.41		mg/m3	Chlorophyll a
Car-4	6/23/2011	WT	6.41		mg/m3	Chlorophyll a
Car-4	8/25/2011	WT	38.45		mg/m3	Chlorophyll a
Car-1	4/27/2011	WT	ND		ug/L	Chlorpyrifos
Car-1	6/23/2011	WT	ND		ug/L	Chlorpyrifos
Car-1	8/25/2011	WT	ND		ug/L	Chlorpyrifos
Car-12	4/27/2011	WT	ND		ug/L	Chlorpyrifos
Car-12	6/23/2011	WT	ND		ug/L	Chlorpyrifos
Car-12	8/25/2011	WT	ND		ug/L	Chlorpyrifos
Car-15	4/27/2011	WT	ND		ug/L	Chlorpyrifos
Car-15	6/23/2011	WT	ND		ug/L	Chlorpyrifos
Car-2	4/27/2011	WT	ND		ug/L	Chlorpyrifos
Car-2	6/23/2011	WT	ND		ug/L	Chlorpyrifos
Car-2	8/25/2011	WT	ND		ug/L	Chlorpyrifos
Car-4	4/27/2011	WT	ND		ug/L	Chlorpyrifos
Car-4	6/23/2011	WT	ND		ug/L	Chlorpyrifos
Car-4	8/25/2011	WT	ND		ug/L	Chlorpyrifos
Car-9	4/27/2011	WT	ND		ug/L	Chlorpyrifos
Car-9	6/23/2011	WT	ND		ug/L	Chlorpyrifos
Car-9	8/25/2011	WT	ND		ug/L	Chlorpyrifos
Car-1	4/27/2011	WT	ND		ug/L	Cyanazine
Car-1	6/23/2011	WT	ND		ug/L	Cyanazine
Car-1	8/25/2011	WT	ND		ug/L	Cyanazine
Car-12	4/27/2011	WT	ND		ug/L	Cyanazine
Car-12	6/23/2011	WT	ND		ug/L	Cyanazine
Car-12	8/25/2011	WT	ND		ug/L	Cyanazine
Car-15	4/27/2011	WT	ND		ug/L	Cyanazine
Car-15	6/23/2011	WT	ND		ug/L	Cyanazine

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-2	4/27/2011	WT	ND		ug/L	Cyanizine
Car-2	6/23/2011	WT	ND		ug/L	Cyanizine
Car-2	8/25/2011	WT	ND		ug/L	Cyanizine
Car-4	4/27/2011	WT	ND		ug/L	Cyanizine
Car-4	6/23/2011	WT	ND		ug/L	Cyanizine
Car-4	8/25/2011	WT	ND		ug/L	Cyanizine
Car-9	4/27/2011	WT	ND		ug/L	Cyanizine
Car-9	6/23/2011	WT	ND		ug/L	Cyanizine
Car-9	8/25/2011	WT	ND		ug/L	Cyanizine
Car-1	4/27/2011	WT	0.64		mg/L	Iron, Total
Car-1	6/23/2011	WT	0.49		mg/L	Iron, Total
Car-1	8/25/2011	WT	0.43		mg/L	Iron, Total
Car-2-10	4/27/2011	WT	0.67		mg/L	Iron, Total
Car-2-10	8/25/2011	WT	0.55		mg/L	Iron, Total
Car-1	4/27/2011	WT	0.10		mg/L	Manganese, Total
Car-1	6/23/2011	WT	0.13		mg/L	Manganese, Total
Car-1	8/25/2011	WT	0.18		mg/L	Manganese, Total
Car-2-10	4/27/2011	WT	0.10		mg/L	Manganese, Total
Car-2-10	8/25/2011	WT	0.20		mg/L	Manganese, Total
Car-1	4/27/2011	WT	ND		ug/L	Metolachlor
Car-1	6/23/2011	WT	1.90		ug/L	Metolachlor
Car-1	8/25/2011	WT	ND		ug/L	Metolachlor
Car-12	4/27/2011	WT	ND		ug/L	Metolachlor
Car-12	6/23/2011	WT	1.60		ug/L	Metolachlor
Car-12	8/25/2011	WT	ND		ug/L	Metolachlor
Car-15	4/27/2011	WT	ND		ug/L	Metolachlor
Car-15	6/23/2011	WT	1.60		ug/L	Metolachlor
Car-2	4/27/2011	WT	ND		ug/L	Metolachlor
Car-2	6/23/2011	WT	ND		ug/L	Metolachlor
Car-2	8/25/2011	WT	ND		ug/L	Metolachlor
Car-4	4/27/2011	WT	ND		ug/L	Metolachlor
Car-4	6/23/2011	WT	ND		ug/L	Metolachlor
Car-4	8/25/2011	WT	ND		ug/L	Metolachlor
Car-9	4/27/2011	WT	ND		ug/L	Metolachlor
Car-9	6/23/2011	WT	ND		ug/L	Metolachlor
Car-9	8/25/2011	WT	ND		ug/L	Metolachlor
Car-1	4/27/2011	WT	ND		ug/L	Metribuzin
Car-1	6/23/2011	WT	ND		ug/L	Metribuzin
Car-1	8/25/2011	WT	ND		ug/L	Metribuzin
Car-12	4/27/2011	WT	ND		ug/L	Metribuzin
Car-12	6/23/2011	WT	ND		ug/L	Metribuzin
Car-12	8/25/2011	WT	ND		ug/L	Metribuzin
Car-15	4/27/2011	WT	ND		ug/L	Metribuzin
Car-15	6/23/2011	WT	ND		ug/L	Metribuzin
Car-2	4/27/2011	WT	ND		ug/L	Metribuzin
Car-2	6/23/2011	WT	ND		ug/L	Metribuzin
Car-2	8/25/2011	WT	ND		ug/L	Metribuzin
Car-4	4/27/2011	WT	ND		ug/L	Metribuzin

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-4	6/23/2011	WT	ND		ug/L	Metribuzin
Car-4	8/25/2011	WT	ND		ug/L	Metribuzin
Car-9	4/27/2011	WT	ND		ug/L	Metribuzin
Car-9	6/23/2011	WT	ND		ug/L	Metribuzin
Car-9	8/25/2011	WT	ND		ug/L	Metribuzin
Car-1	4/27/2011	WT	ND		ug/L	Molinate
Car-1	6/23/2011	WT	ND		ug/L	Molinate
Car-1	8/25/2011	WT	ND		ug/L	Molinate
Car-12	4/27/2011	WT	ND		ug/L	Molinate
Car-12	6/23/2011	WT	ND		ug/L	Molinate
Car-12	8/25/2011	WT	ND		ug/L	Molinate
Car-15	4/27/2011	WT	ND		ug/L	Molinate
Car-15	6/23/2011	WT	ND		ug/L	Molinate
Car-2	4/27/2011	WT	ND		ug/L	Molinate
Car-2	6/23/2011	WT	ND		ug/L	Molinate
Car-2	8/25/2011	WT	ND		ug/L	Molinate
Car-4	4/27/2011	WT	ND		ug/L	Molinate
Car-4	6/23/2011	WT	ND		ug/L	Molinate
Car-4	8/25/2011	WT	ND		ug/L	Molinate
Car-9	4/27/2011	WT	ND		ug/L	Molinate
Car-9	6/23/2011	WT	ND		ug/L	Molinate
Car-9	8/25/2011	WT	ND		ug/L	Molinate
Car-1	4/27/2011	WT	1.10		mg/L	Nitrate-N
Car-1	6/23/2011	WT	0.96		mg/L	Nitrate-N
Car-1	8/25/2011	WT	0.08	J	mg/L	Nitrate-N
Car-12	4/27/2011	WT	1.50		mg/L	Nitrate-N
Car-12	6/23/2011	WT	1.20		mg/L	Nitrate-N
Car-12	8/25/2011	WT	2.40		mg/L	Nitrate-N
Car-15	4/27/2011	WT	1.70		mg/L	Nitrate-N
Car-15	6/23/2011	WT	0.60		mg/L	Nitrate-N
Car-15	8/25/2011	WT	0.50		mg/L	Nitrate-N
Car-2	4/27/2011	WT	1.30		mg/L	Nitrate-N
Car-2	6/23/2011	WT	0.92		mg/L	Nitrate-N
Car-2	8/25/2011	WT	ND		mg/L	Nitrate-N
Car-2-10	4/27/2011	WT	1.30		mg/L	Nitrate-N
Car-2-10	8/25/2011	WT	ND		mg/L	Nitrate-N
Car-4	4/27/2011	WT	1.80		mg/L	Nitrate-N
Car-4	6/23/2011	WT	0.60		mg/L	Nitrate-N
Car-4	8/25/2011	WT	0.48		mg/L	Nitrate-N
Car-9	4/27/2011	WT	1.20		mg/L	Nitrate-N
Car-9	6/23/2011	WT	1.70		mg/L	Nitrate-N
Car-9	8/25/2011	WT	2.70		mg/L	Nitrate-N
Car-1	4/27/2011	WT	0.06		mg/L	Orthophosphate
Car-1	6/23/2011	WT	0.03		mg/L	Orthophosphate
Car-1	8/25/2011	WT	0.19		mg/L	Orthophosphate
Car-12	4/27/2011	WT	0.07		mg/L	Orthophosphate
Car-12	6/23/2011	WT	0.01	J	mg/L	Orthophosphate
Car-12	8/25/2011	WT	0.00	J	mg/L	Orthophosphate

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-15	4/27/2011	WT	0.12		mg/L	Orthophosphate
Car-15	6/23/2011	WT	0.00	J	mg/L	Orthophosphate
Car-15	8/25/2011	WT	0.10	J	mg/L	Orthophosphate
Car-2	4/27/2011	WT	0.04		mg/L	Orthophosphate
Car-2	6/23/2011	WT	0.03		mg/L	Orthophosphate
Car-2	8/25/2011	WT	0.15		mg/L	Orthophosphate
Car-2-10	4/27/2011	WT	0.04		mg/L	Orthophosphate
Car-2-10	8/25/2011	WT	0.17		mg/L	Orthophosphate
Car-4	4/27/2011	WT	0.10		mg/L	Orthophosphate
Car-4	6/23/2011	WT	0.01	J	mg/L	Orthophosphate
Car-4	8/25/2011	WT	0.10	J	mg/L	Orthophosphate
Car-9	4/27/2011	WT	0.09		mg/L	Orthophosphate
Car-9	6/23/2011	WT	0.01	J	mg/L	Orthophosphate
Car-9	8/25/2011	WT	0.01	J	mg/L	Orthophosphate
Car-1	4/27/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-1	6/23/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-1	8/25/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-12	4/27/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-12	6/23/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-12	8/25/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-15	4/27/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-15	6/23/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-2	4/27/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-2	6/23/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-2	8/25/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-4	4/27/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-4	6/23/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-4	8/25/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-9	4/27/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-9	6/23/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-9	8/25/2011	WT	ND		ug/L	Pendimethalin(Prowl)
Car-12	4/27/2011	WT	1.00	<	mg/m3	Pheophytin
Car-12	6/23/2011	WT	1.00	<	mg/m3	Pheophytin
Car-12	8/25/2011	WT	1.00	<	mg/m3	Pheophytin
Car-15	4/27/2011	WT	1.00	<	mg/m3	Pheophytin
Car-15	6/23/2011	WT	1.00	<	mg/m3	Pheophytin
Car-15	8/25/2011	WT	1.00	<	mg/m3	Pheophytin
Car-2	4/27/2011	WT	1.00	<	mg/m3	Pheophytin
Car-2	6/23/2011	WT	1.00	<	mg/m3	Pheophytin
Car-2	8/25/2011	WT	1.00	<	mg/m3	Pheophytin
Car-4	4/27/2011	WT	1.00	<	mg/m3	Pheophytin
Car-4	6/23/2011	WT	1.00	<	mg/m3	Pheophytin
Car-4	8/25/2011	WT	1.00	<	mg/m3	Pheophytin
Car-1	4/27/2011	WT	0.05	J	mg/L	Phosphorus, Total
Car-1	6/23/2011	WT	0.12		mg/L	Phosphorus, Total
Car-1	8/25/2011	WT	0.19		mg/L	Phosphorus, Total
Car-12	4/27/2011	WT	0.07	J	mg/L	Phosphorus, Total
Car-12	6/23/2011	WT	0.11		mg/L	Phosphorus, Total

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-12	8/25/2011	WT	0.03	J	mg/L	Phosphorus, Total
Car-15	4/27/2011	WT	0.10		mg/L	Phosphorus, Total
Car-15	6/23/2011	WT	0.13		mg/L	Phosphorus, Total
Car-15	8/25/2011	WT	0.12		mg/L	Phosphorus, Total
Car-2	4/27/2011	WT	ND		mg/L	Phosphorus, Total
Car-2	6/23/2011	WT	0.13		mg/L	Phosphorus, Total
Car-2	8/25/2011	WT	0.17		mg/L	Phosphorus, Total
Car-2-10	4/27/2011	WT	ND		mg/L	Phosphorus, Total
Car-2-10	8/25/2011	WT	0.17		mg/L	Phosphorus, Total
Car-4	4/27/2011	WT	0.08	J	mg/L	Phosphorus, Total
Car-4	6/23/2011	WT	0.12		mg/L	Phosphorus, Total
Car-4	8/25/2011	WT	0.13		mg/L	Phosphorus, Total
Car-9	4/27/2011	WT	0.06	J	mg/L	Phosphorus, Total
Car-9	6/23/2011	WT	0.12		mg/L	Phosphorus, Total
Car-9	8/25/2011	WT	0.03	J	mg/L	Phosphorus, Total
Car-1	4/27/2011	WT	ND		ug/L	Prometon
Car-1	6/23/2011	WT	ND		ug/L	Prometon
Car-1	8/25/2011	WT	ND		ug/L	Prometon
Car-12	4/27/2011	WT	ND		ug/L	Prometon
Car-12	6/23/2011	WT	ND		ug/L	Prometon
Car-12	8/25/2011	WT	ND		ug/L	Prometon
Car-15	4/27/2011	WT	ND		ug/L	Prometon
Car-15	6/23/2011	WT	ND		ug/L	Prometon
Car-2	4/27/2011	WT	ND		ug/L	Prometon
Car-2	6/23/2011	WT	ND		ug/L	Prometon
Car-2	8/25/2011	WT	ND		ug/L	Prometon
Car-4	4/27/2011	WT	ND		ug/L	Prometon
Car-4	6/23/2011	WT	ND		ug/L	Prometon
Car-4	8/25/2011	WT	ND		ug/L	Prometon
Car-9	4/27/2011	WT	ND		ug/L	Prometon
Car-9	6/23/2011	WT	ND		ug/L	Prometon
Car-9	8/25/2011	WT	ND		ug/L	Prometon
Car-1	4/27/2011	WT	ND		ug/L	Propachlor
Car-1	6/23/2011	WT	ND		ug/L	Propachlor
Car-1	8/25/2011	WT	ND		ug/L	Propachlor
Car-12	4/27/2011	WT	ND		ug/L	Propachlor
Car-12	6/23/2011	WT	ND		ug/L	Propachlor
Car-12	8/25/2011	WT	ND		ug/L	Propachlor
Car-15	4/27/2011	WT	ND		ug/L	Propachlor
Car-15	6/23/2011	WT	ND		ug/L	Propachlor
Car-2	4/27/2011	WT	ND		ug/L	Propachlor
Car-2	6/23/2011	WT	ND		ug/L	Propachlor
Car-2	8/25/2011	WT	ND		ug/L	Propachlor
Car-4	4/27/2011	WT	ND		ug/L	Propachlor
Car-4	6/23/2011	WT	ND		ug/L	Propachlor
Car-4	8/25/2011	WT	ND		ug/L	Propachlor
Car-9	4/27/2011	WT	ND		ug/L	Propachlor
Car-9	6/23/2011	WT	ND		ug/L	Propachlor

Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-9	8/25/2011	WT	ND		ug/L	Propachlor
Car-1	4/27/2011	WT	0.38	J	ug/L	Simazine
Car-1	6/23/2011	WT	ND		ug/L	Simazine
Car-1	8/25/2011	WT	ND		ug/L	Simazine
Car-12	4/27/2011	WT	0.26	J	ug/L	Simazine
Car-12	6/23/2011	WT	ND		ug/L	Simazine
Car-12	8/25/2011	WT	ND		ug/L	Simazine
Car-15	4/27/2011	WT	ND		ug/L	Simazine
Car-15	6/23/2011	WT	ND		ug/L	Simazine
Car-2	4/27/2011	WT	0.69	J	ug/L	Simazine
Car-2	6/23/2011	WT	ND		ug/L	Simazine
Car-2	8/25/2011	WT	ND		ug/L	Simazine
Car-4	4/27/2011	WT	ND		ug/L	Simazine
Car-4	6/23/2011	WT	ND		ug/L	Simazine
Car-4	8/25/2011	WT	ND		ug/L	Simazine
Car-9	4/27/2011	WT	ND		ug/L	Simazine
Car-9	6/23/2011	WT	ND		ug/L	Simazine
Car-9	8/25/2011	WT	ND		ug/L	Simazine
Car-1	4/27/2011	WT	5.90		mg/L	Total Organic Carbon (TOC)
Car-1	6/23/2011	WT	5.60		mg/L	Total Organic Carbon (TOC)
Car-1	8/25/2011	WT	6.50		mg/L	Total Organic Carbon (TOC)
Car-12	4/27/2011	WT	8.20		mg/L	Total Organic Carbon (TOC)
Car-12	6/23/2011	WT	8.00		mg/L	Total Organic Carbon (TOC)
Car-12	8/25/2011	WT	4.50		mg/L	Total Organic Carbon (TOC)
Car-15	4/27/2011	WT	9.60		mg/L	Total Organic Carbon (TOC)
Car-15	6/23/2011	WT	7.90		mg/L	Total Organic Carbon (TOC)
Car-15	8/25/2011	WT	5.90		mg/L	Total Organic Carbon (TOC)
Car-2	4/27/2011	WT	6.10		mg/L	Total Organic Carbon (TOC)
Car-2	6/23/2011	WT	5.20		mg/L	Total Organic Carbon (TOC)
Car-2	8/25/2011	WT	5.60		mg/L	Total Organic Carbon (TOC)
Car-2-10	4/27/2011	WT	6.20		mg/L	Total Organic Carbon (TOC)
Car-2-10	8/25/2011	WT	5.90		mg/L	Total Organic Carbon (TOC)
Car-4	4/27/2011	WT	8.40		mg/L	Total Organic Carbon (TOC)
Car-4	6/23/2011	WT	7.70		mg/L	Total Organic Carbon (TOC)
Car-4	8/25/2011	WT	6.20		mg/L	Total Organic Carbon (TOC)
Car-9	4/27/2011	WT	8.30		mg/L	Total Organic Carbon (TOC)
Car-9	6/23/2011	WT	7.90		mg/L	Total Organic Carbon (TOC)
Car-9	8/25/2011	WT	4.40		mg/L	Total Organic Carbon (TOC)
Car-1	4/27/2011	WT	30.00		mg/L	Total Suspended Solids
Car-1	6/23/2011	WT	21.00		mg/L	Total Suspended Solids
Car-1	8/25/2011	WT	17.00		mg/L	Total Suspended Solids
Car-12	4/27/2011	WT	93.00		mg/L	Total Suspended Solids
Car-12	6/23/2011	WT	113.00		mg/L	Total Suspended Solids
Car-12	8/25/2011	WT	75.00		mg/L	Total Suspended Solids
Car-15	4/27/2011	WT	63.00		mg/L	Total Suspended Solids
Car-15	6/23/2011	WT	60.00		mg/L	Total Suspended Solids
Car-15	8/25/2011	WT	32.00		mg/L	Total Suspended Solids
Car-2	4/27/2011	WT	31.00		mg/L	Total Suspended Solids



Site	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Car-2	6/23/2011	WT	34.00		mg/L	Total Suspended Solids
Car-2	8/25/2011	WT	19.00		mg/L	Total Suspended Solids
Car-2-10	4/27/2011	WT	31.00		mg/L	Total Suspended Solids
Car-2-10	8/25/2011	WT	15.00		mg/L	Total Suspended Solids
Car-4	4/27/2011	WT	61.00		mg/L	Total Suspended Solids
Car-4	6/23/2011	WT	57.00		mg/L	Total Suspended Solids
Car-4	8/25/2011	WT	21.00		mg/L	Total Suspended Solids
Car-9	4/27/2011	WT	142.00		mg/L	Total Suspended Solids
Car-9	6/23/2011	WT	261.00		mg/L	Total Suspended Solids
Car-9	8/25/2011	WT	61.00		mg/L	Total Suspended Solids
Car-1	4/27/2011	WT	ND		ug/L	Trifluralin
Car-1	6/23/2011	WT	ND		ug/L	Trifluralin
Car-1	8/25/2011	WT	ND		ug/L	Trifluralin
Car-12	4/27/2011	WT	ND		ug/L	Trifluralin
Car-12	6/23/2011	WT	ND		ug/L	Trifluralin
Car-12	8/25/2011	WT	ND		ug/L	Trifluralin
Car-15	4/27/2011	WT	ND		ug/L	Trifluralin
Car-15	6/23/2011	WT	ND		ug/L	Trifluralin
Car-2	4/27/2011	WT	ND		ug/L	Trifluralin
Car-2	6/23/2011	WT	ND		ug/L	Trifluralin
Car-2	8/25/2011	WT	ND		ug/L	Trifluralin
Car-4	4/27/2011	WT	ND		ug/L	Trifluralin
Car-4	6/23/2011	WT	ND		ug/L	Trifluralin
Car-4	8/25/2011	WT	ND		ug/L	Trifluralin
Car-9	4/27/2011	WT	ND		ug/L	Trifluralin
Car-9	6/23/2011	WT	ND		ug/L	Trifluralin
Car-9	8/25/2011	WT	ND		ug/L	Trifluralin
Car-1	4/27/2011	WT	11.00		mg/L	Volatile Suspended Solids
Car-1	6/23/2011	WT			mg/L	Volatile Suspended Solids
Car-1	8/25/2011	WT	ND		mg/L	Volatile Suspended Solids
Car-12	4/27/2011	WT	19.00		mg/L	Volatile Suspended Solids
Car-12	6/23/2011	WT	8.00		mg/L	Volatile Suspended Solids
Car-12	8/25/2011	WT	6.00		mg/L	Volatile Suspended Solids
Car-15	4/27/2011	WT	14.00		mg/L	Volatile Suspended Solids
Car-15	6/23/2011	WT	11.00		mg/L	Volatile Suspended Solids
Car-15	8/25/2011	WT	9.00		mg/L	Volatile Suspended Solids
Car-2	4/27/2011	WT	13.00		mg/L	Volatile Suspended Solids
Car-2	6/23/2011	WT	7.00		mg/L	Volatile Suspended Solids
Car-2	8/25/2011	WT	ND		mg/L	Volatile Suspended Solids
Car-2-10	4/27/2011	WT	12.00		mg/L	Volatile Suspended Solids
Car-2-10	8/25/2011	WT	ND		mg/L	Volatile Suspended Solids
Car-4	4/27/2011	WT	20.00		mg/L	Volatile Suspended Solids
Car-4	6/23/2011	WT	9.00		mg/L	Volatile Suspended Solids
Car-4	8/25/2011	WT	ND		mg/L	Volatile Suspended Solids
Car-9	4/27/2011	WT	26.00		mg/L	Volatile Suspended Solids
Car-9	6/23/2011	WT	13.00		mg/L	Volatile Suspended Solids
Car-9	8/25/2011	WT	6.00		mg/L	Volatile Suspended Solids

U Analyte was not detected

J Estimated value between Method Detection Limit (MDL) and Practical Quantitation Limit (PQL)

### **E. coli DATA**

	Date Collected	Matrix	Result	Qualifier	Unit	Analyte Name
Boulder	4/27/2011	w	789		NPN/100ml	E. Coli
Boulder	8/25/2011	w	10		NPN/100ml	E. Coli
Boulder	6/23/2011	w	186		NPN/100ml	E. Coli
Dam West	4/27/2011	w	169		NPN/100ml	E. Coli
Dam West	8/25/2011	w	10		NPN/100ml	E. Coli
Dam West	6/23/2011	w	109		NPN/100ml	E. Coli
Keysport	4/27/2011	w	173		NPN/100ml	E. Coli
Keysport	8/25/2011	w	10		NPN/100ml	E. Coli
Keysport	6/23/2011	w	41		NPN/100ml	E. Coli

<b>Beach Data</b>		<b>E. coli (mpn/100ml)</b>											
		<b>Keysport</b>											
		<b>April</b>		<b>May</b>		<b>June</b>		<b>July</b>		<b>August</b>		<b>September</b>	
<b>Day</b>		<b>East</b>	<b>West</b>	<b>East</b>	<b>West</b>	<b>East</b>	<b>East</b>	<b>East</b>	<b>East</b>	<b>East</b>	<b>East</b>	<b>East</b>	<b>East</b>
6		0	0	0	1	1	0	1	1	0	0	0	0
13		0	1	0	0	0	0	0	0	4	1	0	0
20		0	0	0	0	0	0	1	1	3	5	0	0
27		0	0	0	0	0	0	4	2	0	1	0	0
29						1	0						
		<b>Dam West</b>											
<b>Day</b>		<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>
6		0	0	0	0	0	0	0	0	0	0	0	0
13		0	0	0	0	0	0	0	0	6	2	0	0
20		0	0	0	0	0	0	0	0	0	0	0	0
27		0	0	0	0	0	0	0	0	0	0	0	0
29						0	0						
		<b>McNair</b>											
<b>Day</b>		<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>
6		0	0	0	0	0	0	0	0	0	0	0	0
13		0	0	0	0	0	0	0	1	0	0	0	0
20		0	0	0	0	0	0	0	0	0	0	0	0
27		0	0	0	0	0	0	0	0	0	0	0	0
29						0	0						
		<b>Coles Cr.</b>											
<b>Day</b>		<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>	<b>North</b>	<b>South</b>
6		0	0	6	3	3	2	3	2	0	0	0	2
13		0	0	0	0	7	5	2	2	8	6	0	0
20		0	0	0	0	2	3	5	5	12	8	0	0
27		0	0	1	1	0	0	4	3	3	5	0	0
29						0	0						
		<b>Boulder (potable)</b>											
<b>Day</b>		<b>North</b>		<b>North</b>		<b>North</b>		<b>North</b>		<b>North</b>		<b>North</b>	
6		0		0		0		0		0		0	
13				0				0		0		0	
20				0				0		0		0	
27				0				0		0		0	
29													

## FIELD DATA

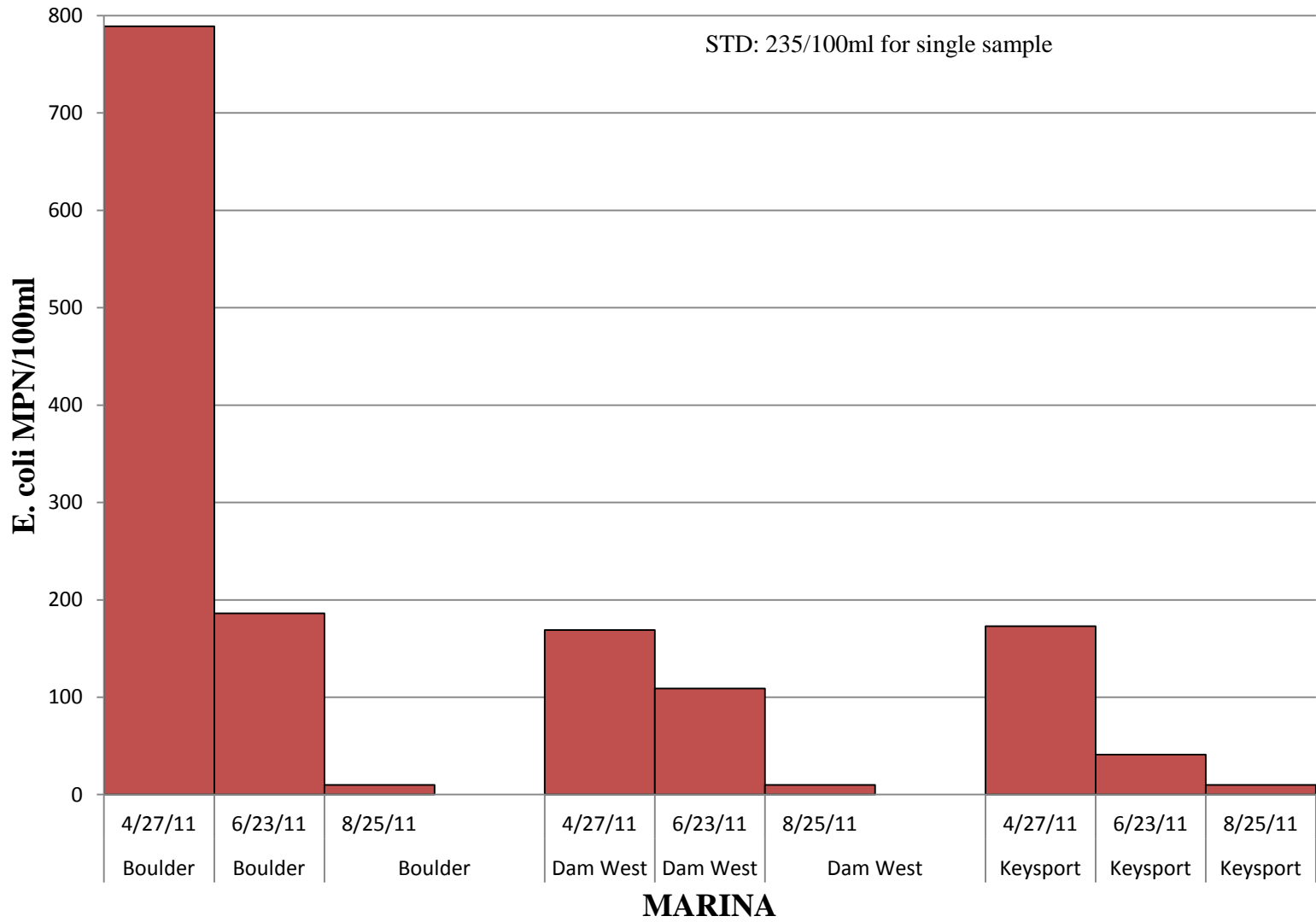
Site	Date	Depth	H2OTemp (oC)	Redox	Cond	D.O.%	D.O. (mg/L)	pH	Time	Secchi (in.)
1	4/27/2011	0.4	16.40	298	404.0	93.0	9.01	8.43	843	
1	6/23/2011	1.2	24.96	319	348.1	107.6	8.80	5.48	841	
1	8/25/2011	1.4	26.96	347	295.0	99.0	7.90	8.24	827	
1	10/27/2011	0.15	5.83	374	314.0	97.1	12.70	5.98		
2	4/27/2011	0.5	16.10	286	412.0	108.7	8.91	5.97	910	15
2	4/27/2011	1.0	16.10	281	412.0	107.8	8.83	6.11		
2	4/27/2011	2.0	16.10	279	411.0	107.5	8.81	6.14		
2	4/27/2011	3.0	16.10	277	411.0	107.0	8.78	6.16		
2	4/27/2011	4.0	16.10	276	411.0	106.9	8.75	6.15		
2	4/27/2011	5.0	16.10	282	410.0	106.7	8.74	6.15		
2	6/23/2011	0.6	24.95	347	349.1	70.0	5.74	5.37	925	10
2	6/23/2011	1.1	24.93	348	349.2	67.1	5.50	5.35	926	
2	6/23/2011	2.1	24.94	348	349.3	67.7	5.53	5.34	927	
2	6/23/2011	3.0	24.93	348	349.5	67.6	5.53	5.32	928	
2	6/23/2011	4.0	24.93	347	349.3	66.0	5.39	5.33	929	
2	6/23/2011	5.0	24.95	346	349.4	66.0	5.43	5.34	930	
2	6/23/2011	6.1	24.94	350	349.3	65.3	5.33	5.31	931	
2	8/25/2011	0.0	27.12	326	293.1	64.6	5.19	8.39	905	
2	8/25/2011	1.0	27.10	327	293.7	62.1	4.95	8.41		12
2	8/25/2011	2.0	27.10	327	293.1	62.4	4.98	8.41		
2	8/25/2011	3.0	27.10	327	293.5	62.3	4.98	8.4		
2	8/25/2011	4.0	27.11	328	293.0	63.1	4.92	8.37		
2	8/25/2011	5.0	27.10	329	294.0	59.7	4.77	8.37		
2	8/25/2011	6.0	27.10	329	293.1	59.8	4.79	8.36		
2	8/25/2011	7.0	27.11	329	294.0	59.2	4.68	8.33		
2	8/25/2011	8.0	27.11	329	293.2	57.5	4.58	8.34		
2	8/25/2011	9.0	27.10	329	293.2	58.0	4.62	8.34		
2	8/25/2011	9.3	27.10	329	293.9	58.1	4.62	8.33		
2	10/27/2011	0.15	13.58	145	321.7	85.2	8.77	8.6		
2	10/27/2011	0.46	13.60	153	322.6	81.6	8.46	8.52		
2	10/27/2011	0.76	13.60	162	323.1	80.8	8.38	8.43		
2	10/27/2011	1.07	13.60	168	322.9	80.4	8.35	8.38		
2	10/27/2011	1.37	13.60	174	322.6	80.3	8.34	8.34		
2	10/27/2011	1.68	13.60	177	322.0	80.0	8.31	8.35		
2	10/27/2011	1.98	13.60	177	322.1	79.8	8.29	8.4		
2	10/27/2011	2.29	13.60	181	322.0	79.6	8.27	8.42		
4	4/27/2011	0.5	16.80	319	282.0	79.1	7.59	7.89	1130	4
4	4/27/2011	1.0	16.80	322	280.0	78.9	7.59	7.81		
4	4/27/2011	2.0	16.80	326	280.0	78.7	7.58	7.8		
4	4/27/2011	3.0	16.80	328	280.0	78.7	7.57	7.8		
4	4/27/2011	4.0	16.80	331	281.0	78.8	7.57	7.8		
4	4/27/2011	5.0	16.80	332	275.0	79.4	7.65	7.8		
4	4/27/2011	6.0	16.80	333	276.0	79.8	7.68	7.7		
4	4/27/2011	7.0	16.80	334	275.0	79.8	7.68	7.8		
4	4/27/2011	8.0	16.80	66	418.0	46.7	2.12	7.5		

4	6/23/2011	0.5	24.38	324	166.0	70.6	5.71	5.02	1215	10
4	8/25/2011	0.0	27.75	302	375.0	84.0	6.46	8.36	1026	10
4	8/25/2011	1.0	27.68	303	374.5	76.0	5.88	8.36		
4	8/25/2011	2.0	27.55	305	376.8	69.3	5.46	8.3		
4	8/25/2011	3.0	27.49	309	381.9	63.6	5.03	8.23		
4	10/27/2011	0.15	13.96	181	323.2	90.9	9.29	8.31		
4	10/27/2011	0.46	13.99	184	322.5	84.7	8.68	8.31		
4	10/27/2011	0.76	13.99	186	322.5	83.0	8.55	8.33		
4	10/27/2011	1.07	13.98	188	323.1	82.7	8.51	8.33		
4	10/27/2011	1.37	14.01	191	324.0	82.0	8.44	8.34		
9	4/27/2011	0.2	16.80	318	224.0	72.7	7.05	7.76	1320	
9	6/23/2011	0.5	22.88	5	340.5	55.5	4.51	4.5	1111	
9	8/25/2011	0.0	26.23	324	408.1	96.4	7.76	8.2	1145	
12	4/27/2011	0.5	16.80	334	216.0	66.6	6.40	7.6	1400	
12	6/23/2011	0.4	24.40	353	221.8	55.2	4.54	4.75	1026	
12	8/25/2011	0.0	27.34	304	415.6	98.0	7.73	8.15	1222	

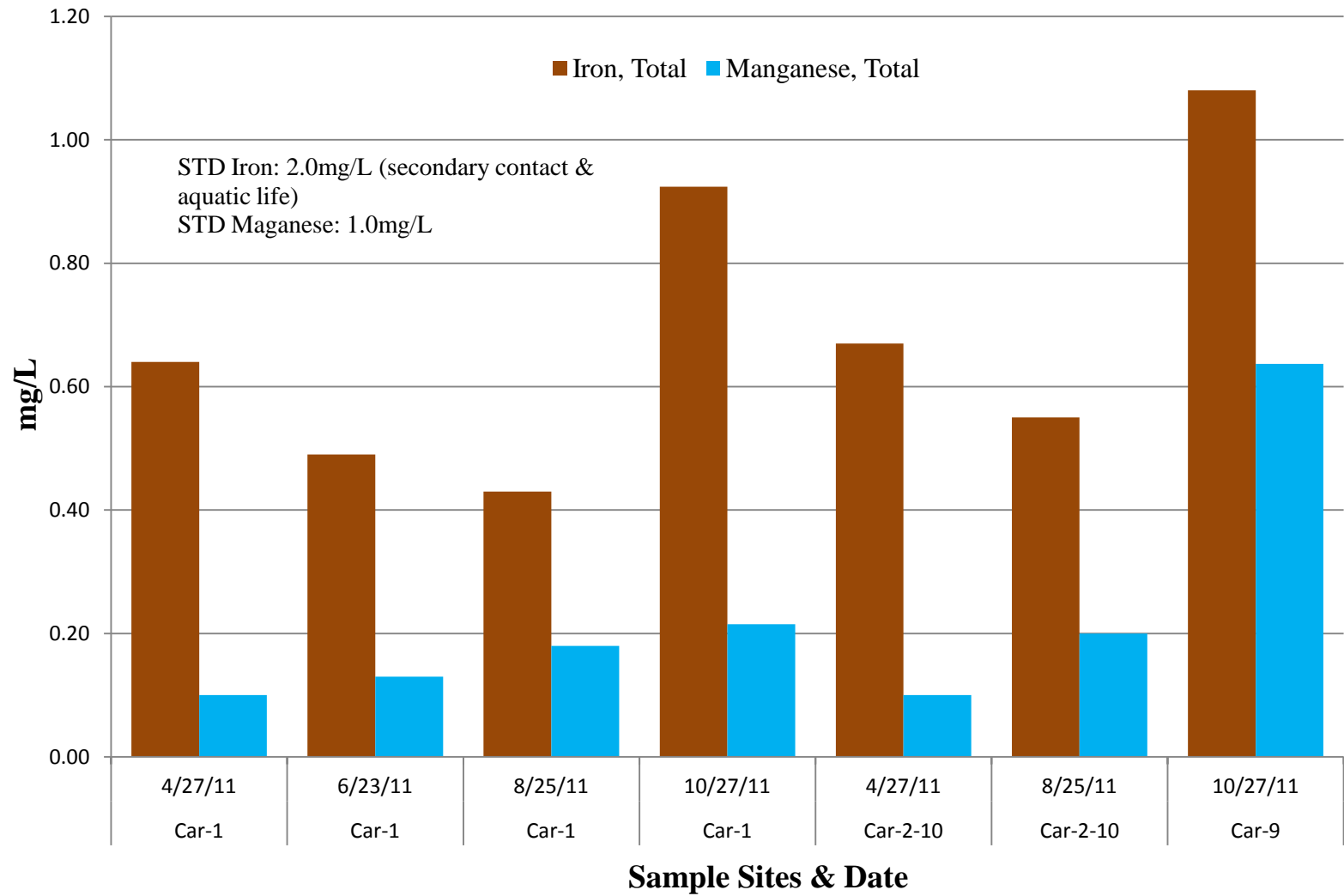
## **APPENDIX B**

### **LAB DATA GRAPHS**

# E. coli at Marinas

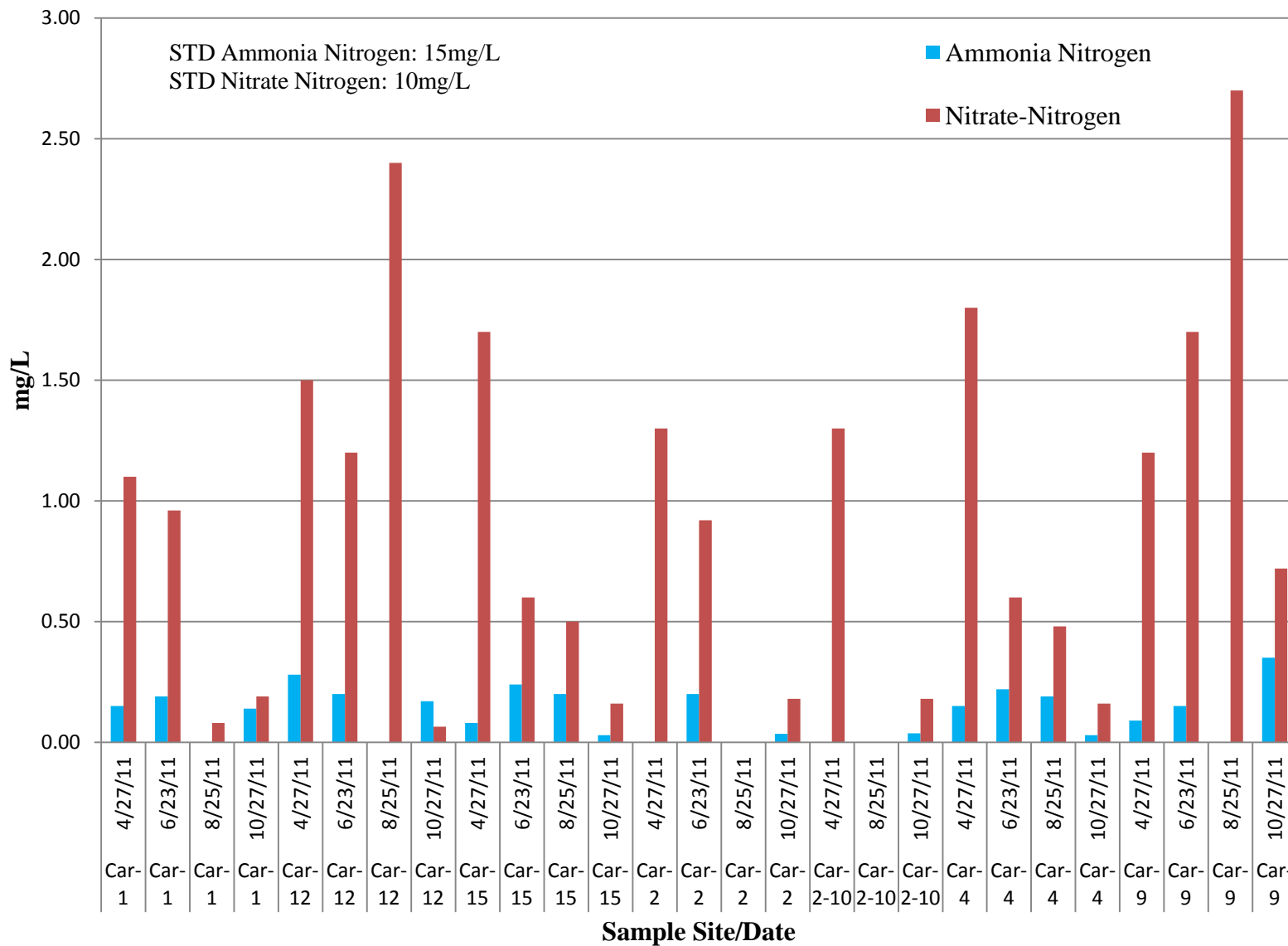


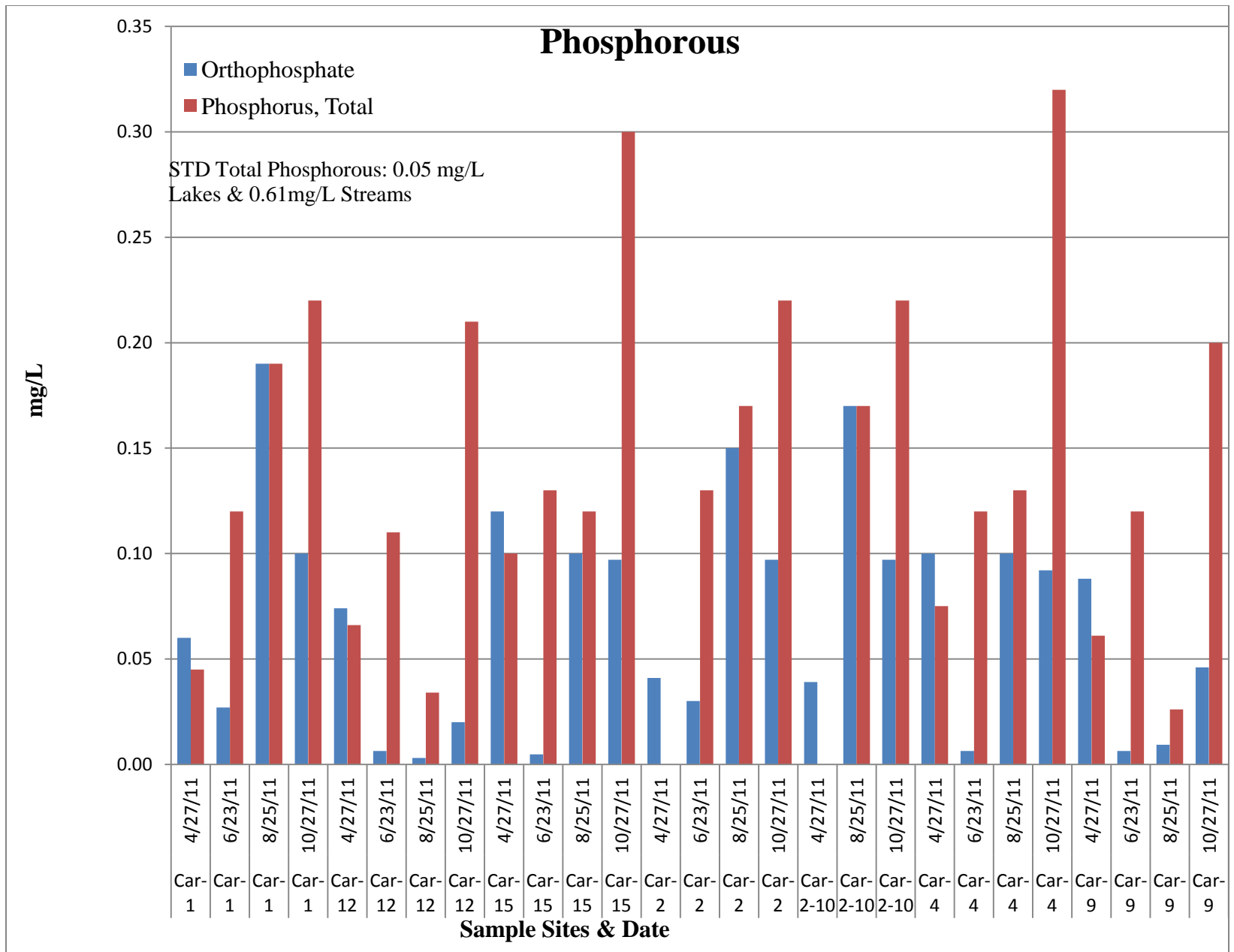
# TOTAL IRON & MAGANESE



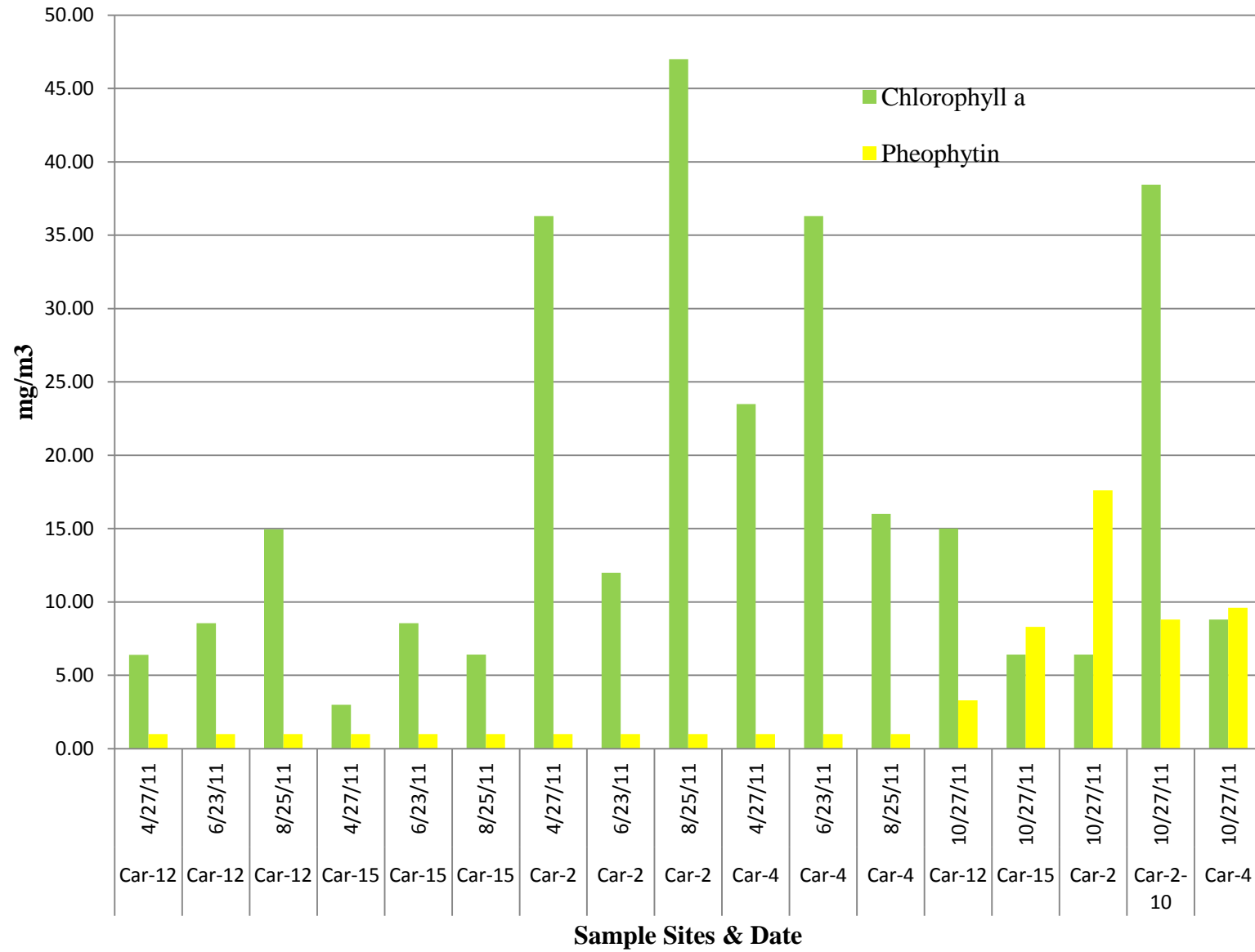


# Nitrogen & Ammonia

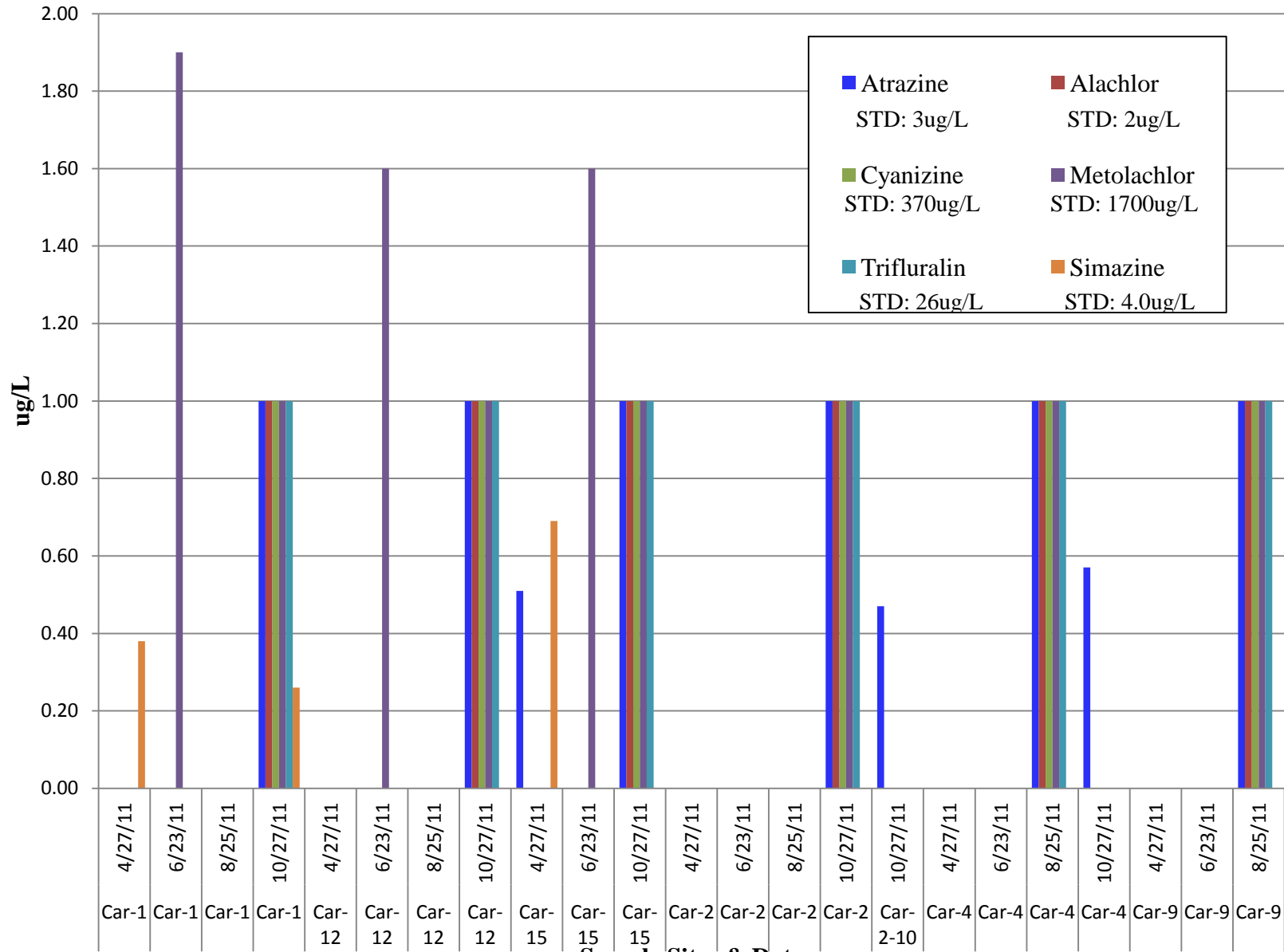




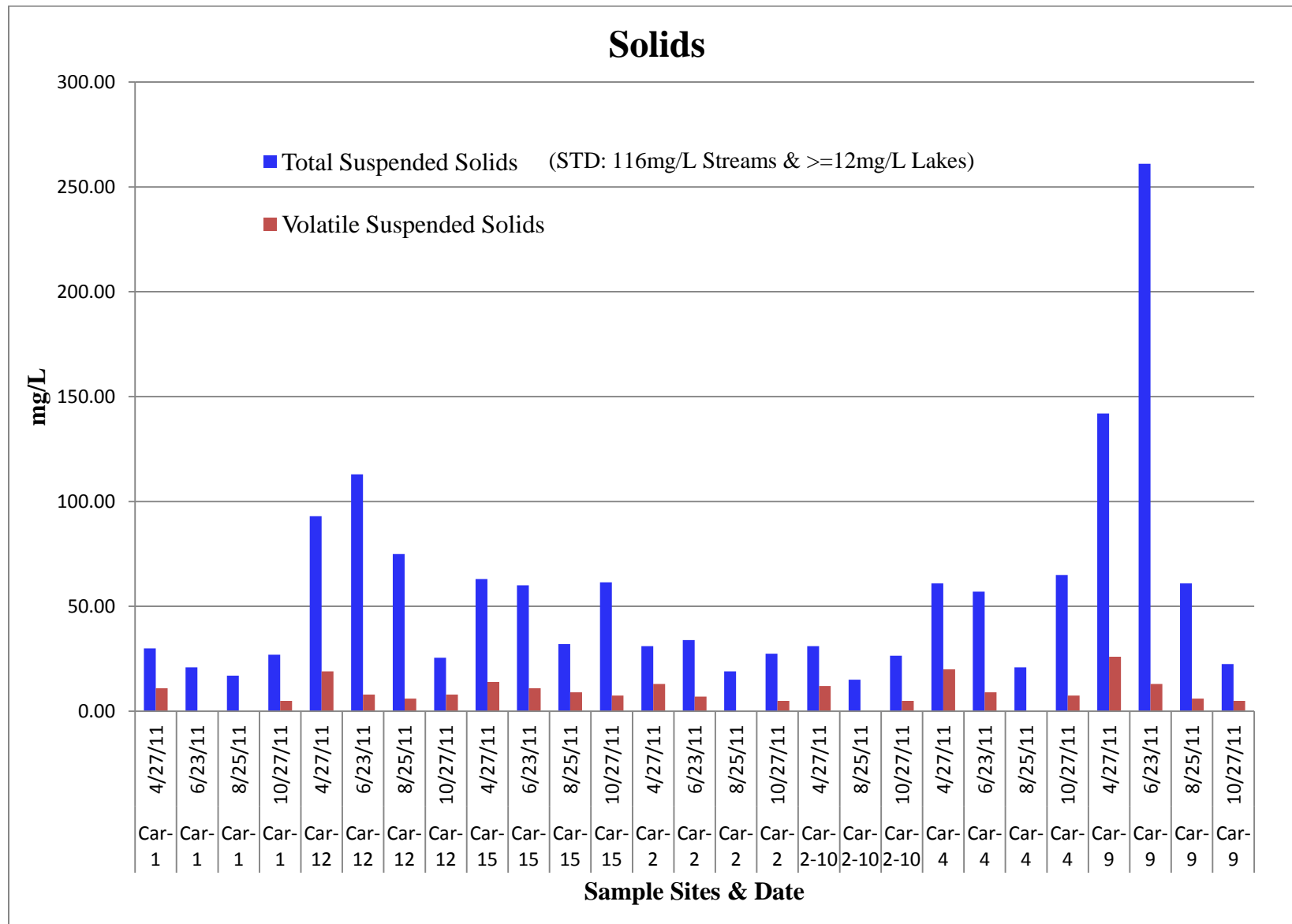
# Chlorophyll & Pheophytin



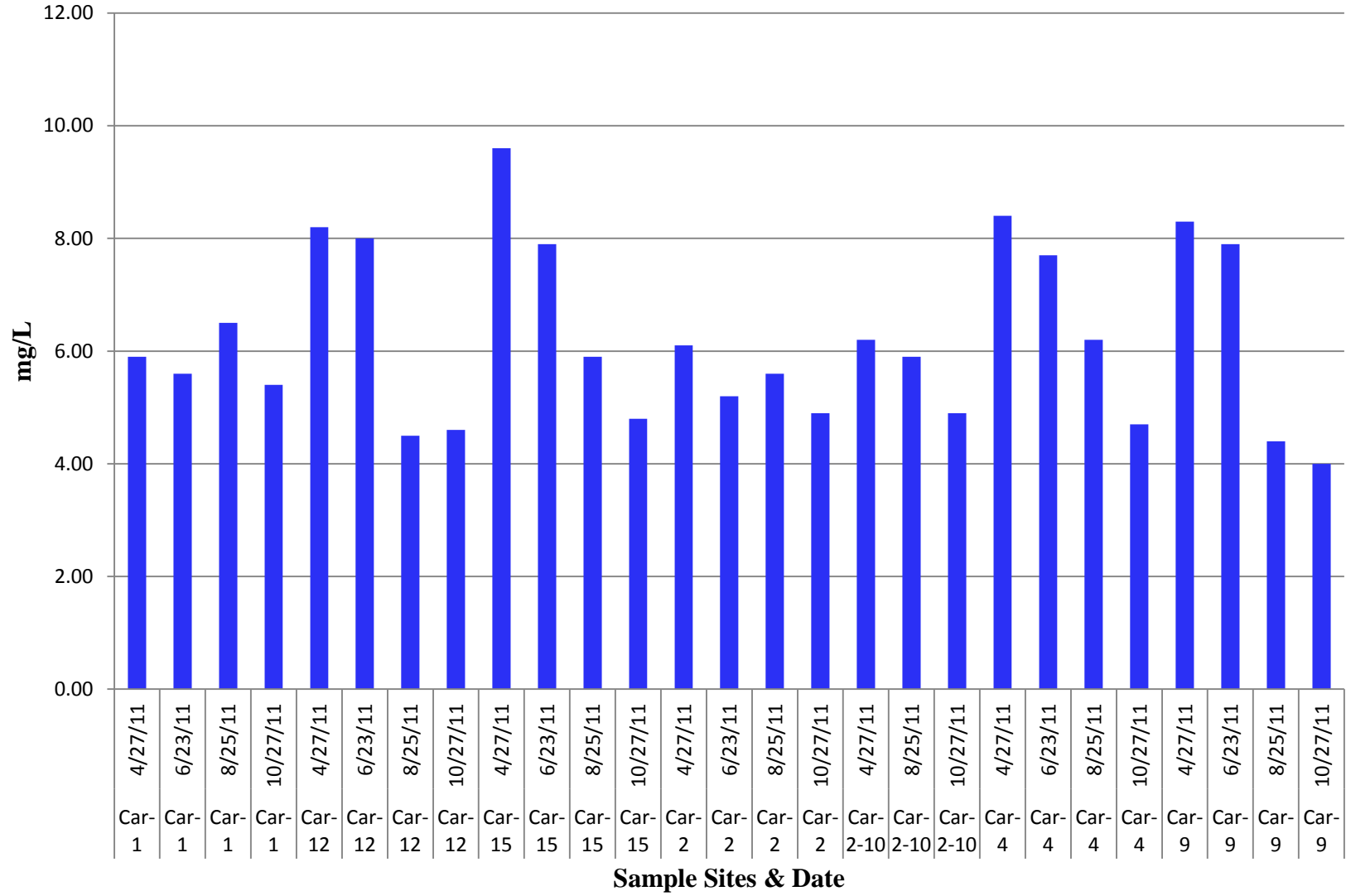
# Pesticides



Sample Sites & Date



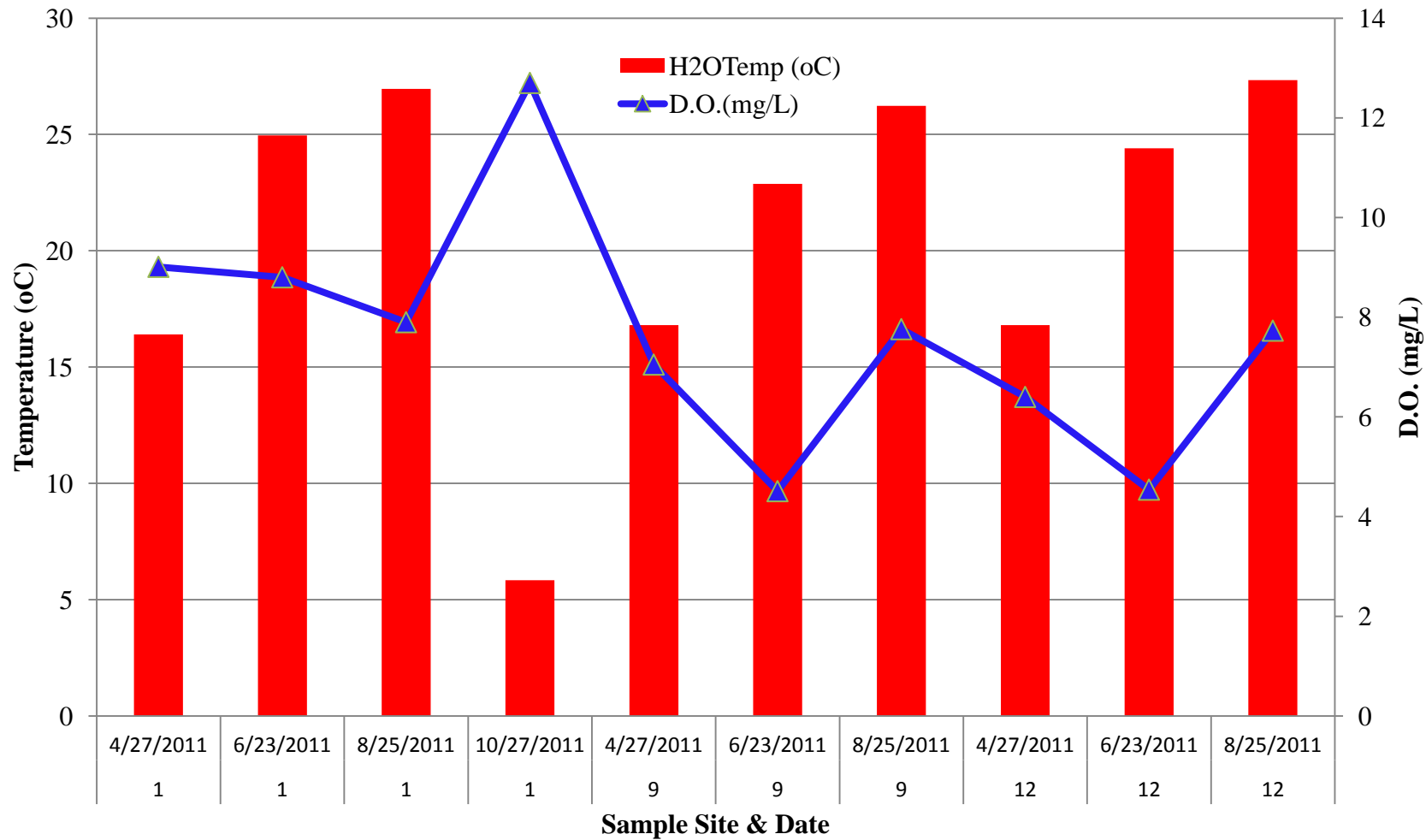
# Total Organic Carbon



## **APPENDIX C**

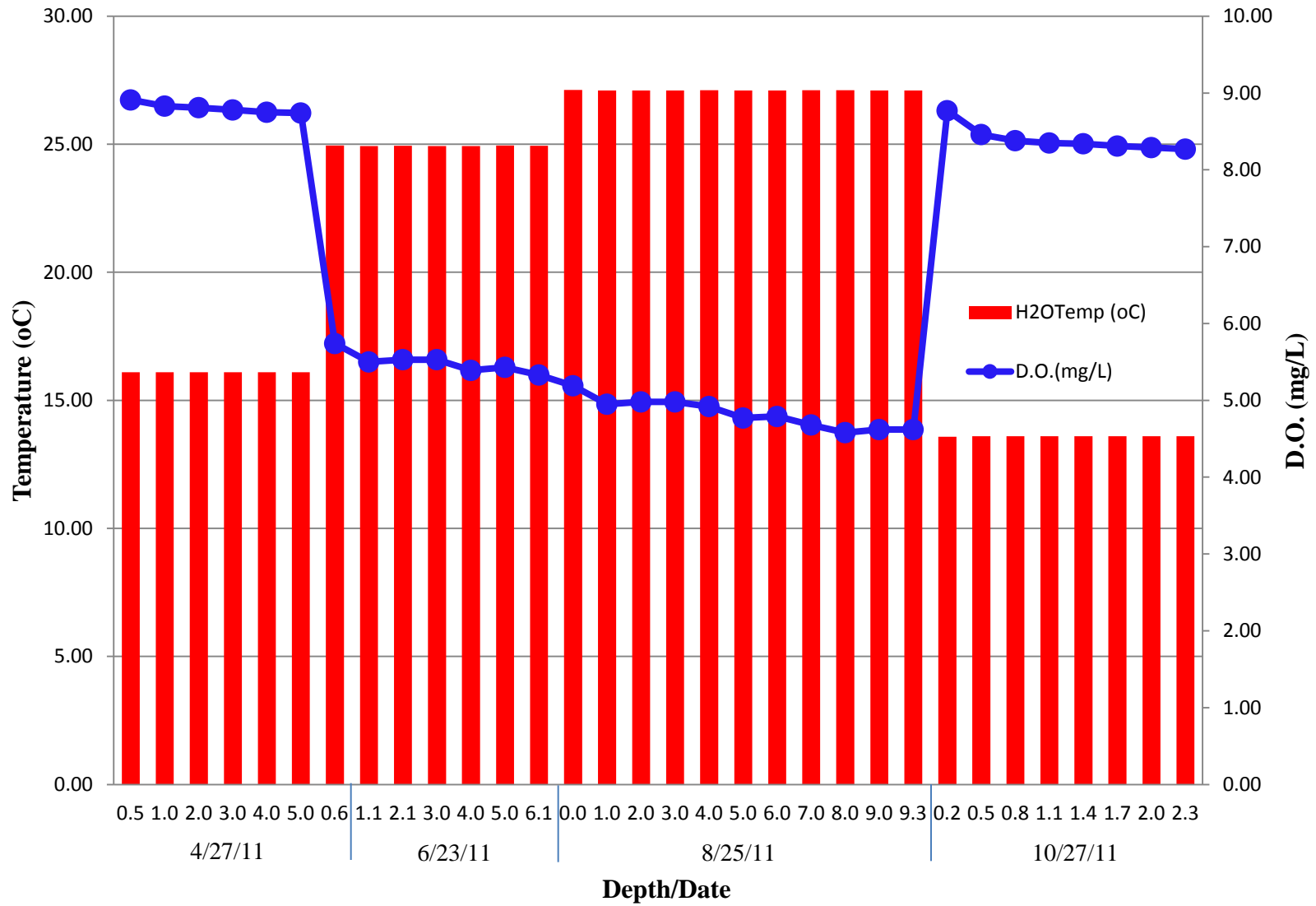
### **FIELD DATA GRAPHS**

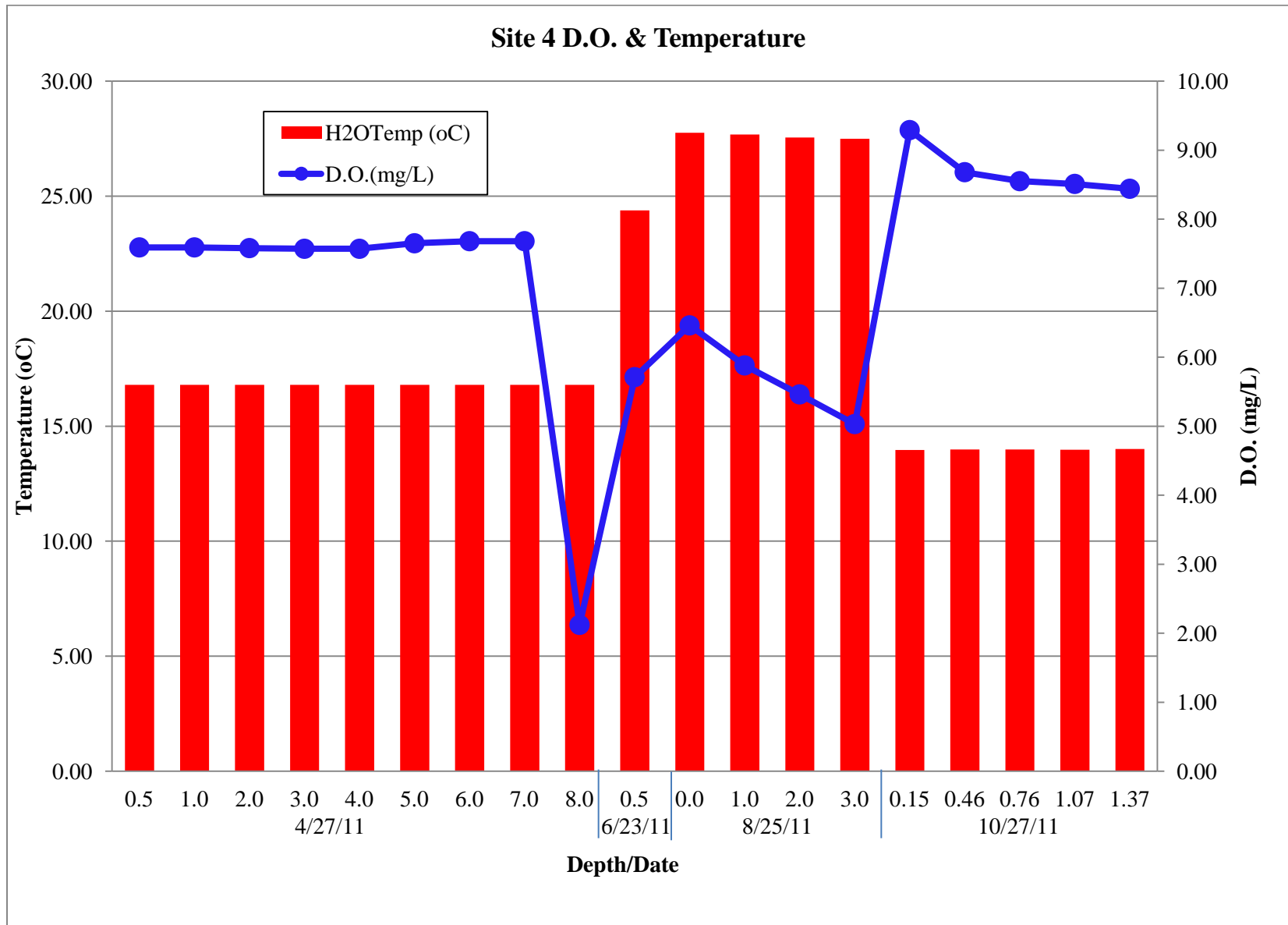
### Tributaries D.O. & Temperatures



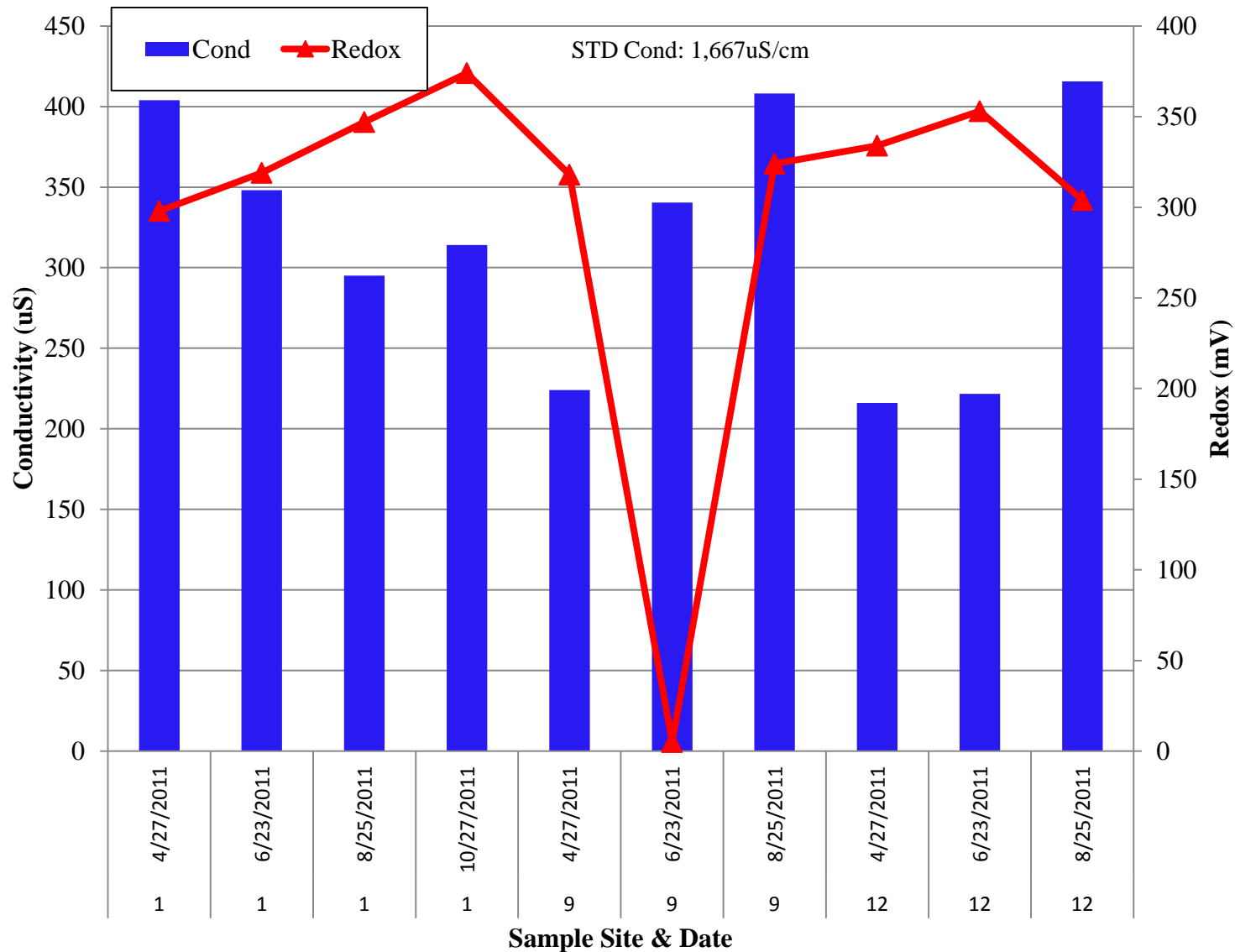


# Site 2 D.O. & Temperature

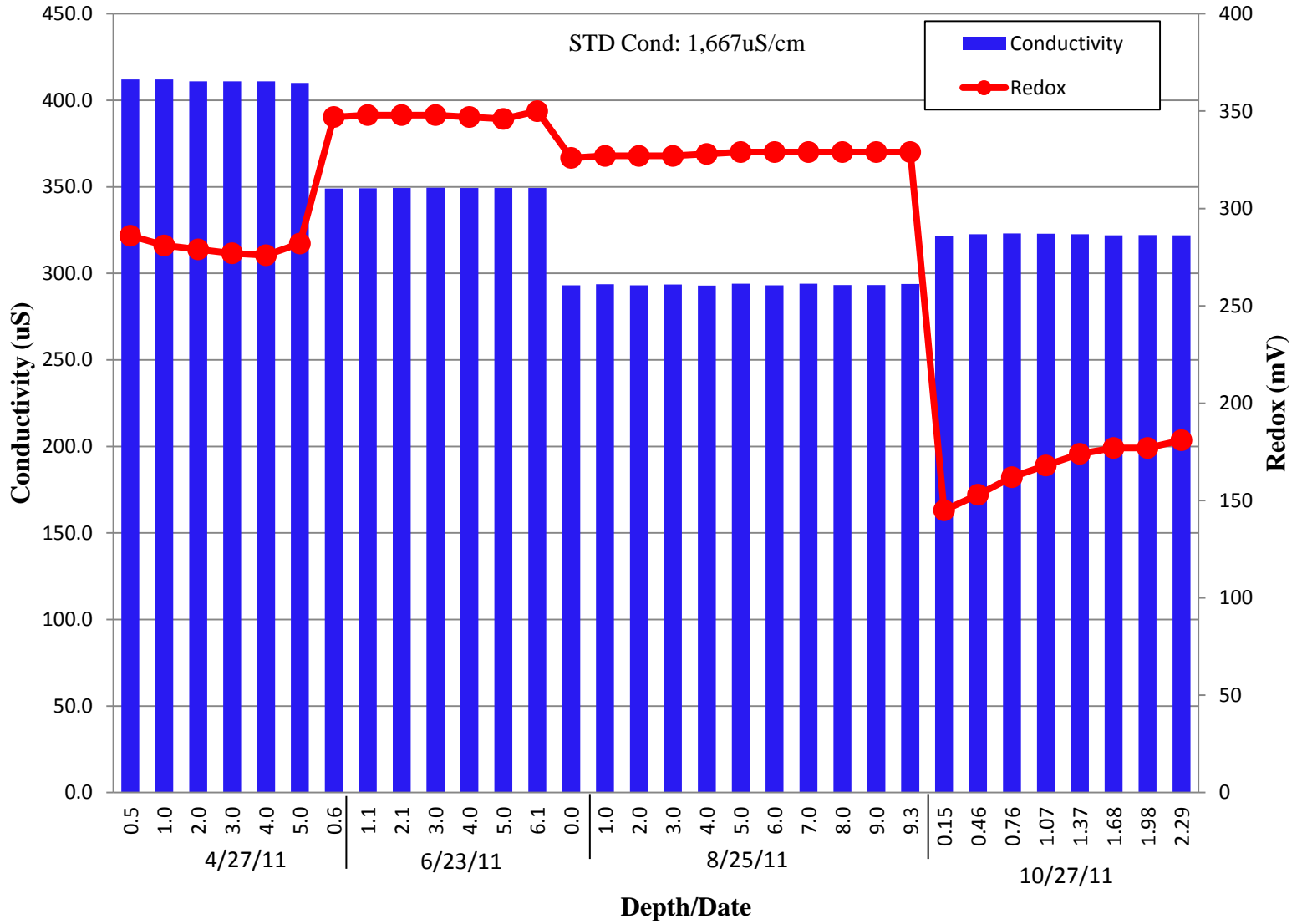




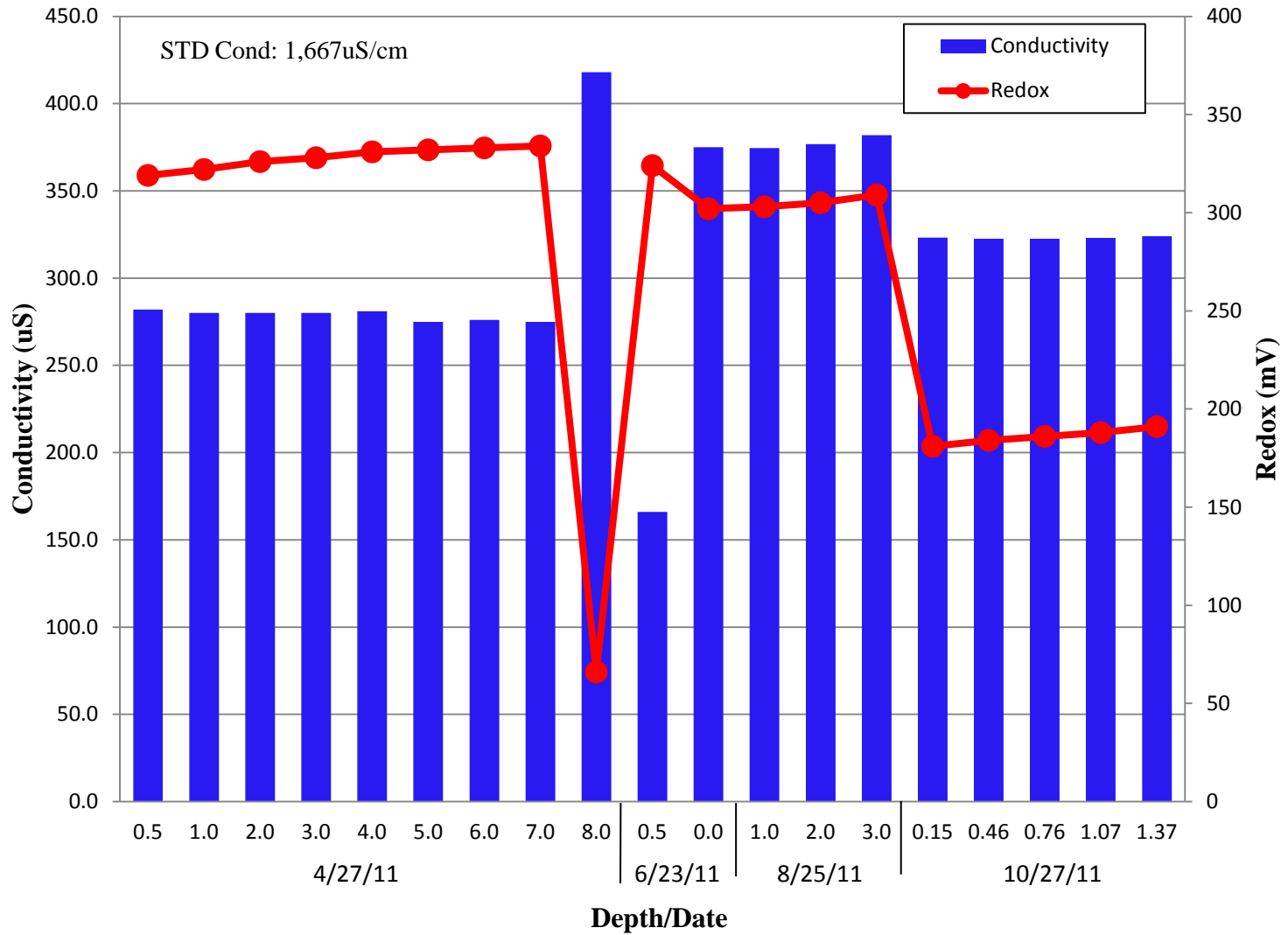
# Tributary Redox & Conductivity



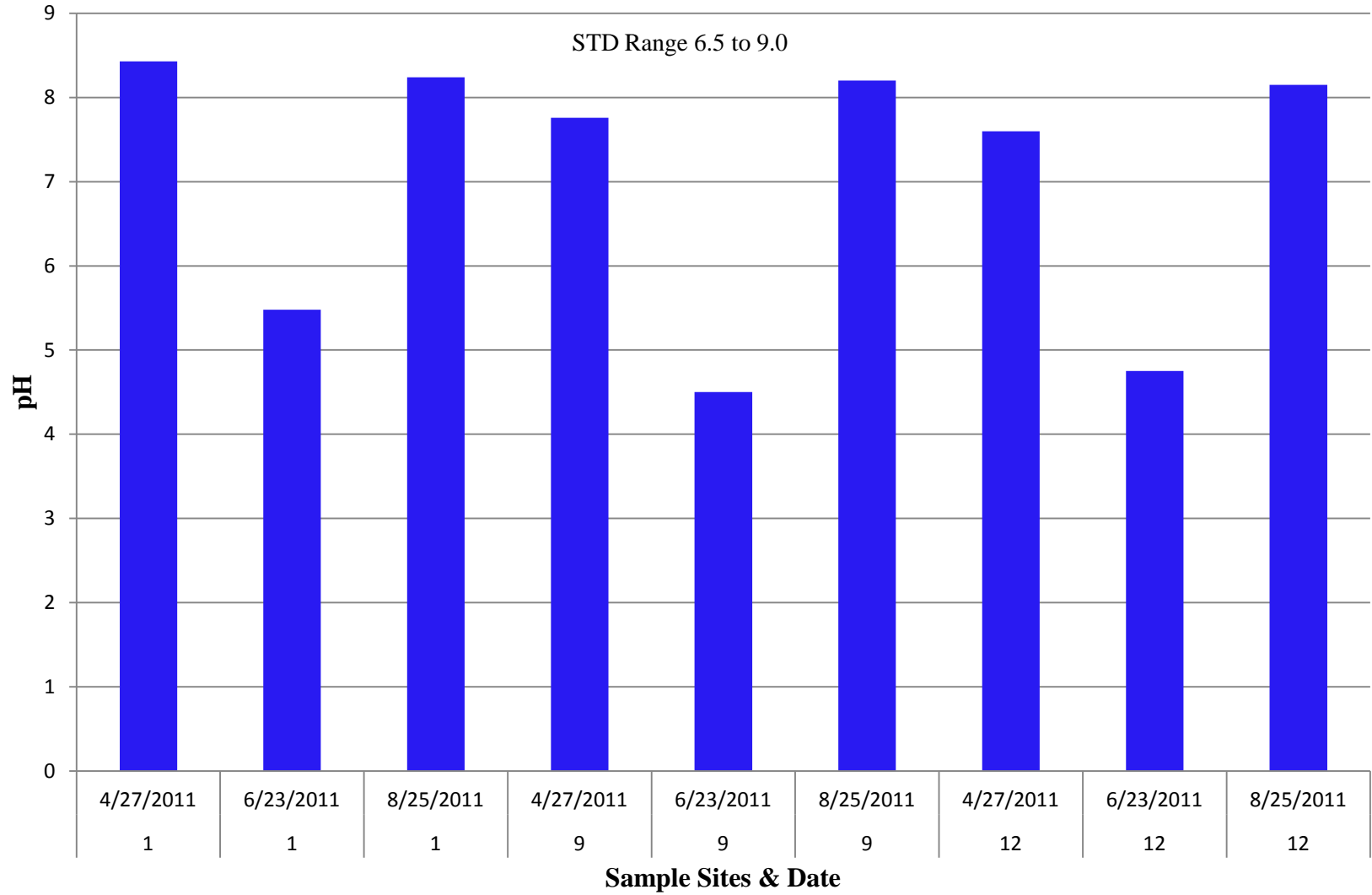
# Site 2 Conductivity & Redox

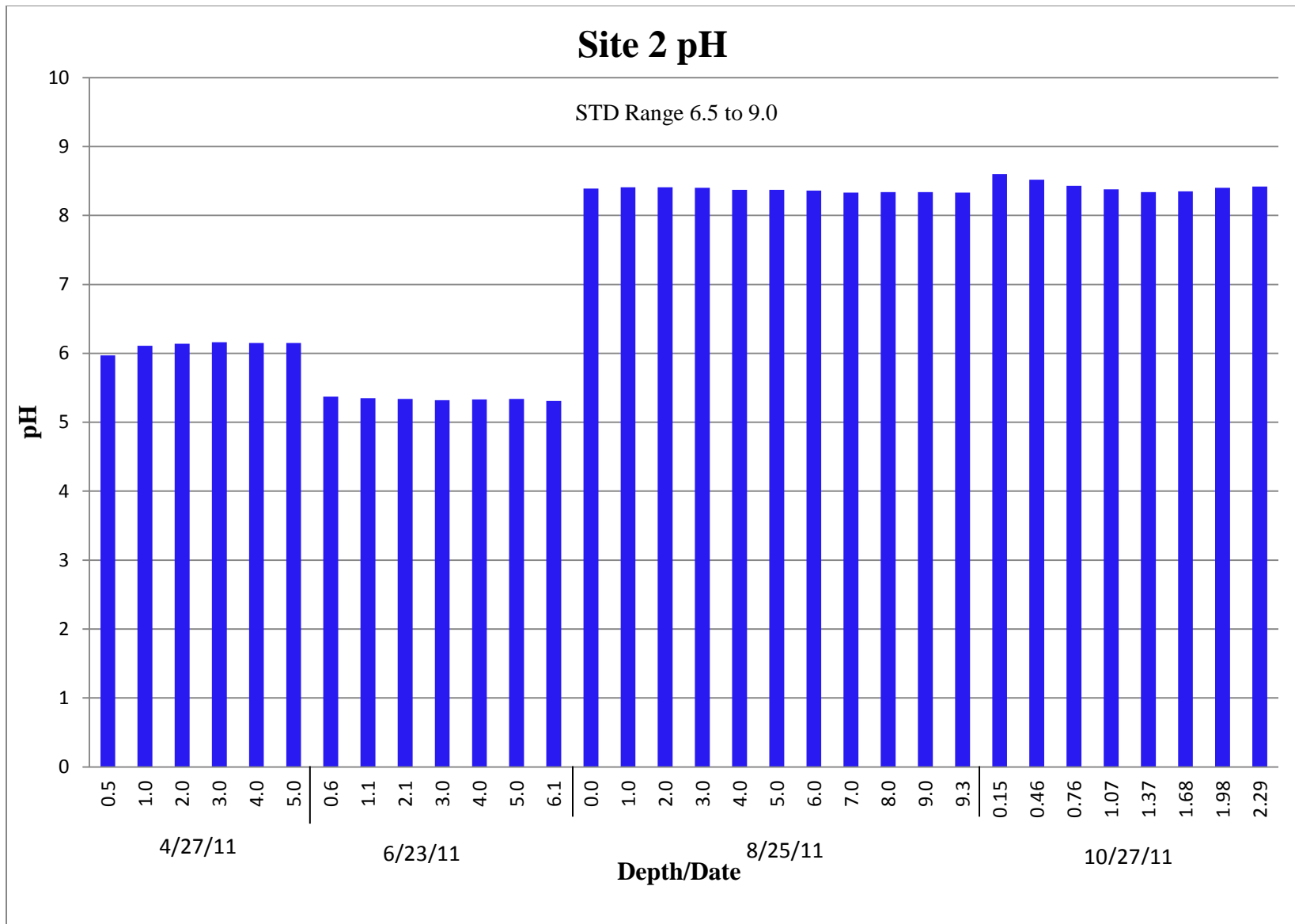


### Site 4 Conductivity & Redox

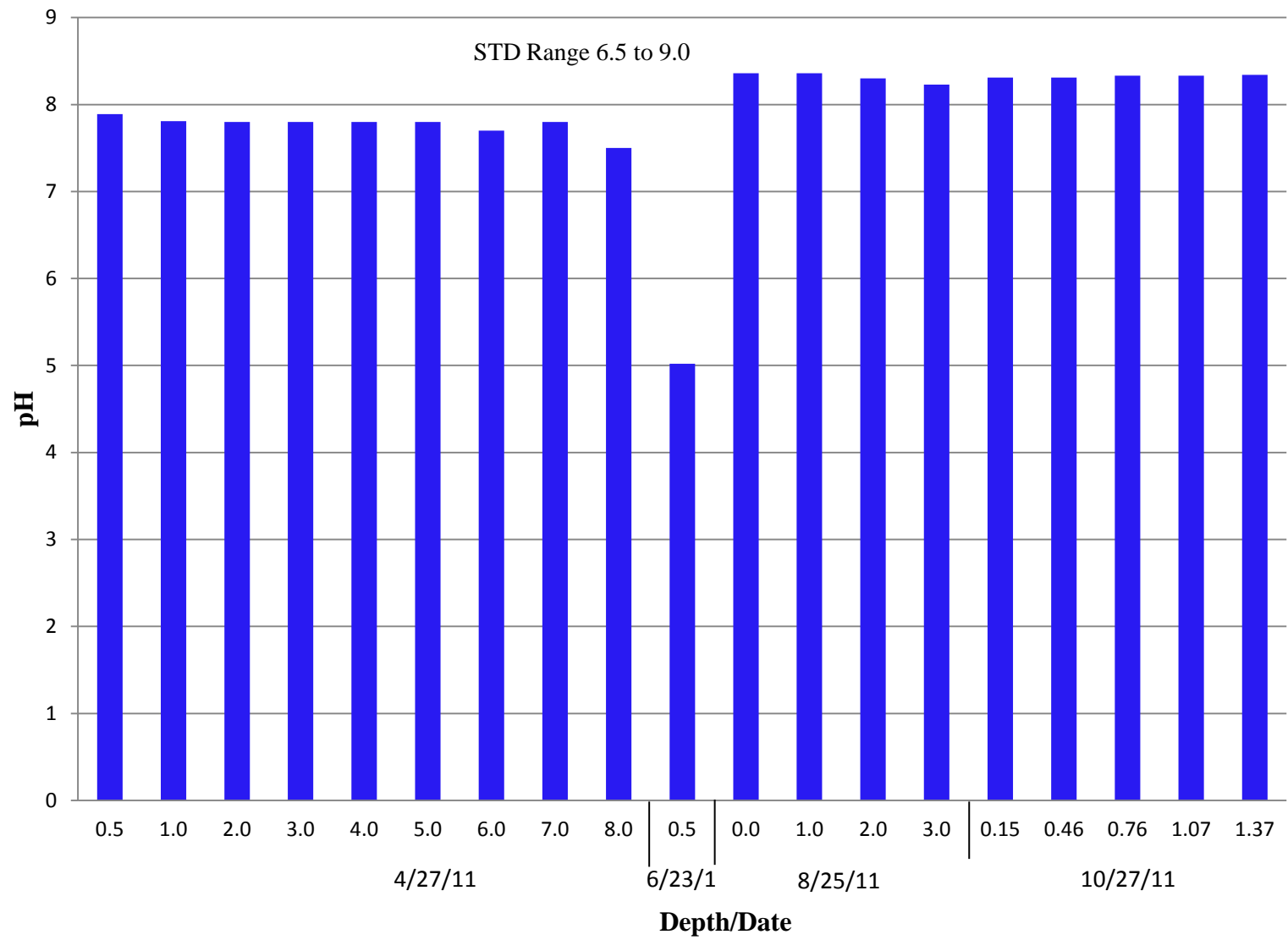


# pH Tributary sites



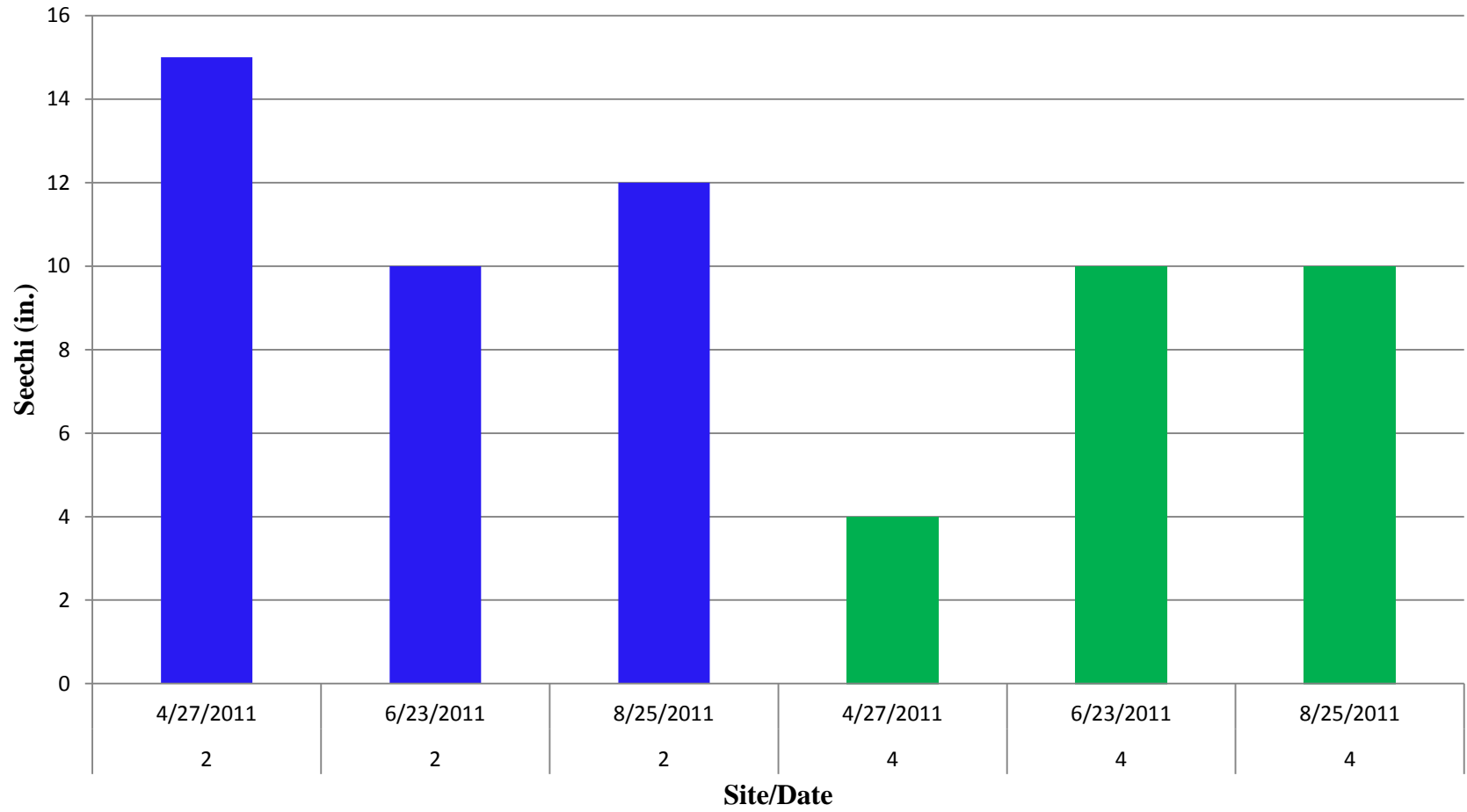


# Site 4 pH





# Secchi

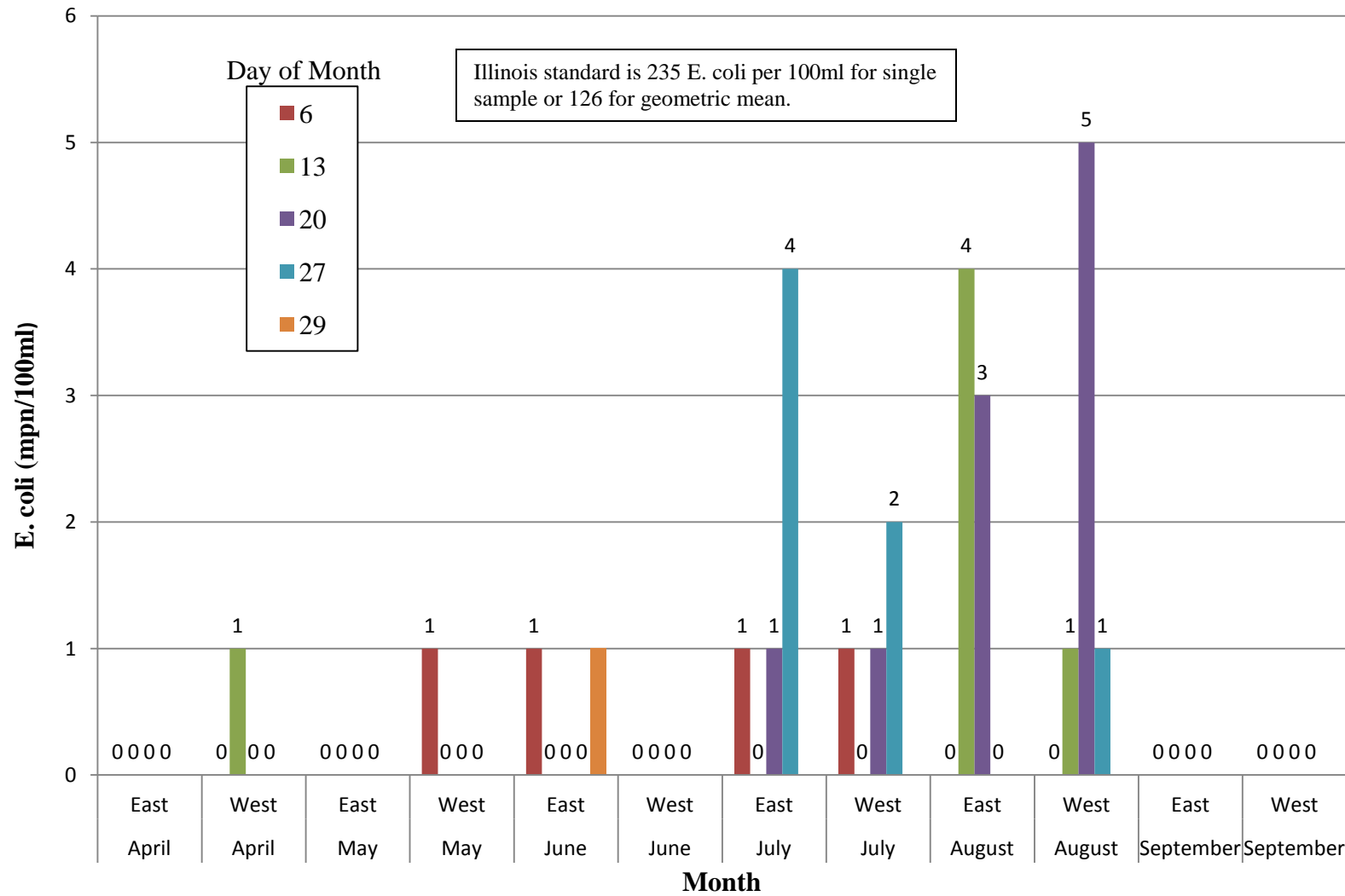




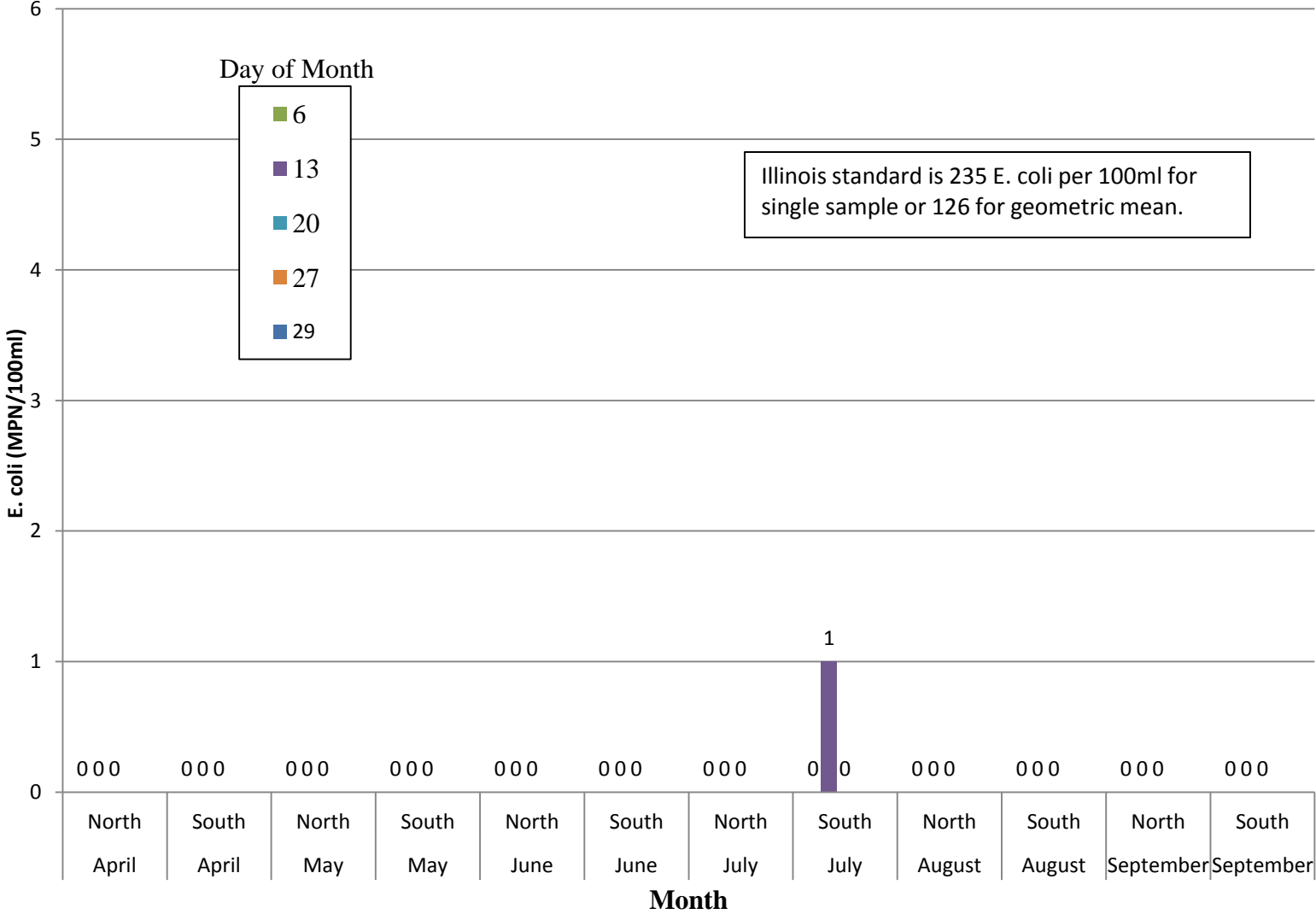
## **APPENDIX D**

### **BEACH GRAPHS**

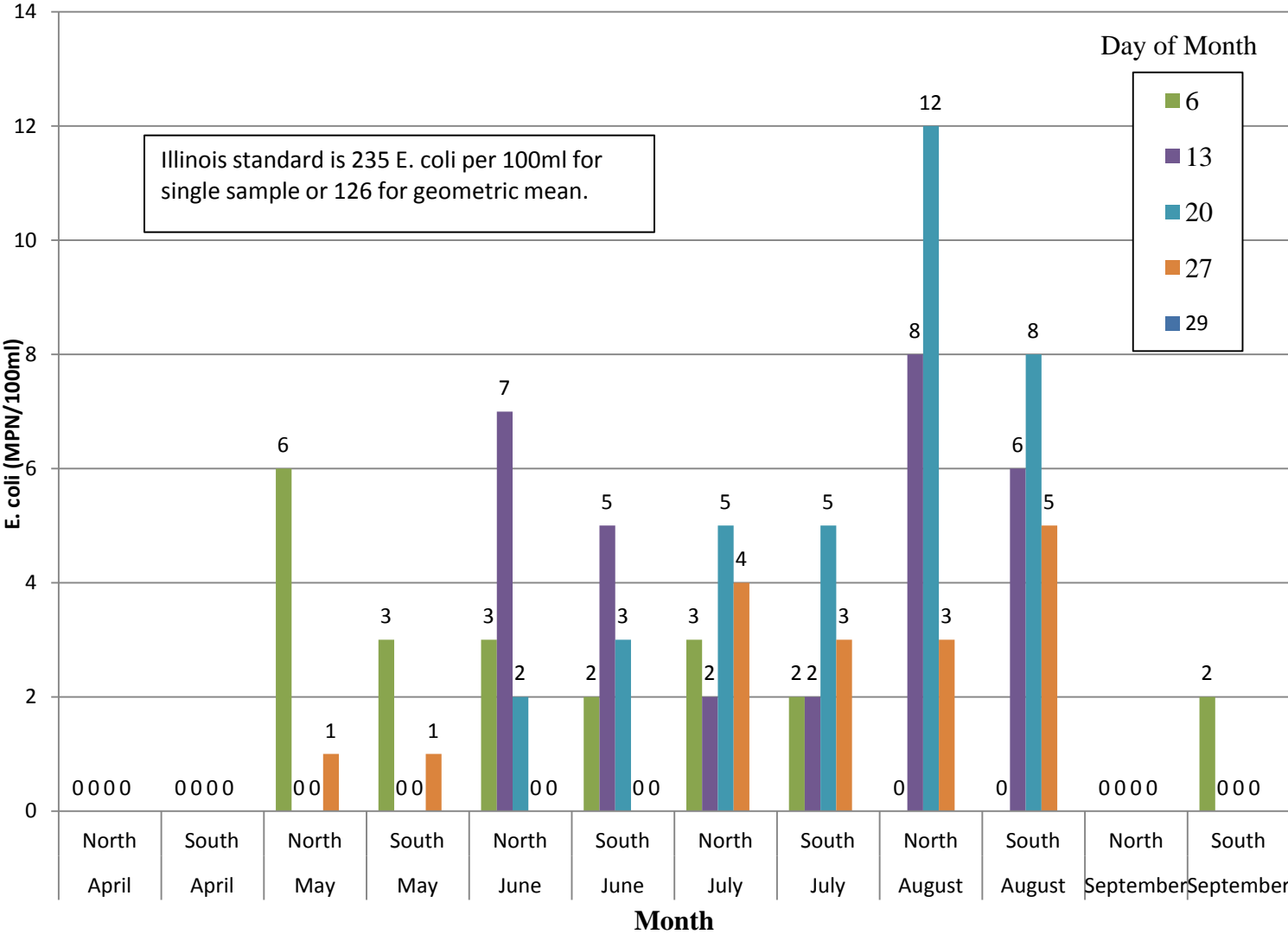
# Keysport Beach



# McNair Beach



# Coles Creek Beach



# Dam West Beach

